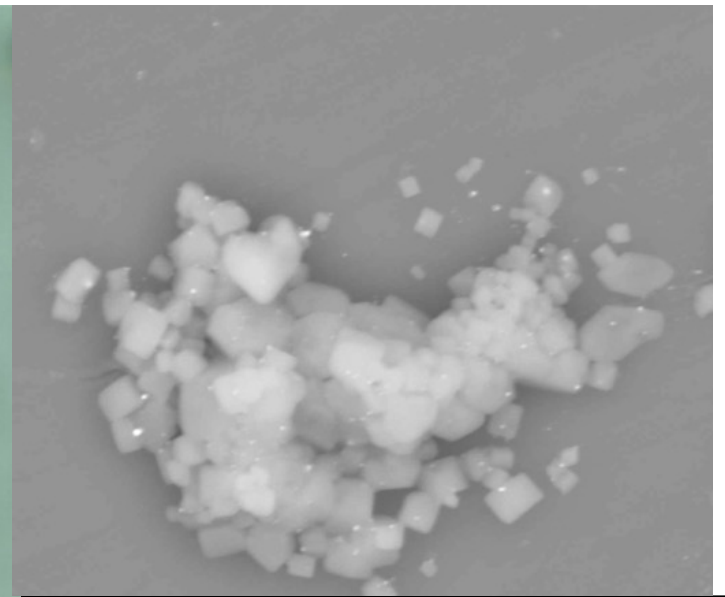




DIVEX workshop 2017



The role and character of fluids in rare-metal deposits: insights from Thor Lake, NWT.

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Department of Earth and
Environmental Sciences



University
of Windsor

References

Much of the data presented in this talk comes from the thesis work of Yonggang Feng and Justin Hoyle:

Feng, Y., 2014, Hydrothermal Geochemistry and Mineralizing Processes in the T Zone, Thor Lake Rare-element Deposit, Northwest Territories. PhD Thesis, University of Windsor. 341 p.

Hoyle, J., 2017, Rare-Earth Elements in the Nechalacho Deposit, NWT: Hydrothermal Controls on Mineralogy and Fractionation, MSc Thesis, University of Windsor. 91 p.

Partly published as:

Feng, Y., Samson, I.M., 2015, Replacement Processes involving high field strength elements in the T Zone, Thor Lake rare-metal deposit, Northwest Territories. *Canadian Mineralogist*. v.53, p.31-60.

The summary diagrams at the end come from :

Samson, I.M., 2013, Fluid inclusion studies of rare earth element deposits (*abstract*). Geological Society of America, 125th Anniversary Meeting, Denver, Colorado, Oct. 27-30.

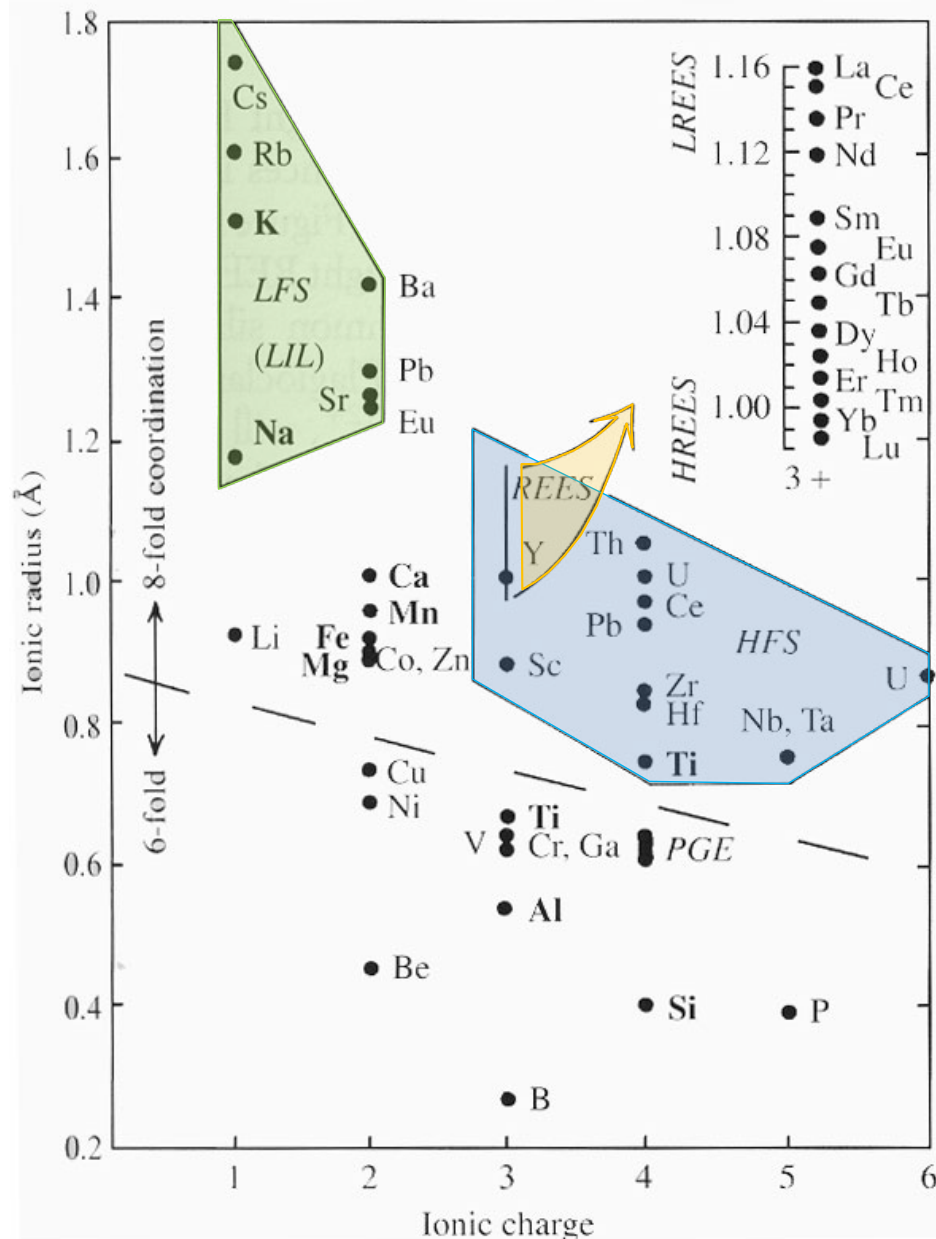
“Rare Metals”

1 H																	2 He
3 Li	4 Be	Lanthanides (Ce-Lu) Lanthanons (La-Lu + Y)										5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Uuu	112 Uub			114 Uuq			

LREE ← | → HREE

58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
Th	91 Pa	U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

Classification of Elements Based on Charge and Radius



Ionic Potential - charge/radius –
low solubility in aqueous solutions?

“immobile”?

High Field Strength (HFS)
– Zr, Nb, Ta, Ti, REE

Can they be mobile?
What is the scale of mobility?

Hydrothermal REE

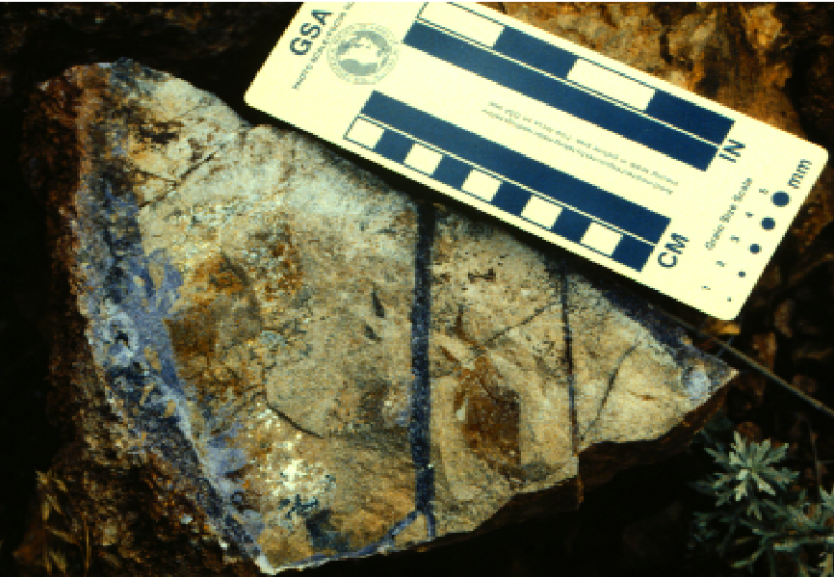
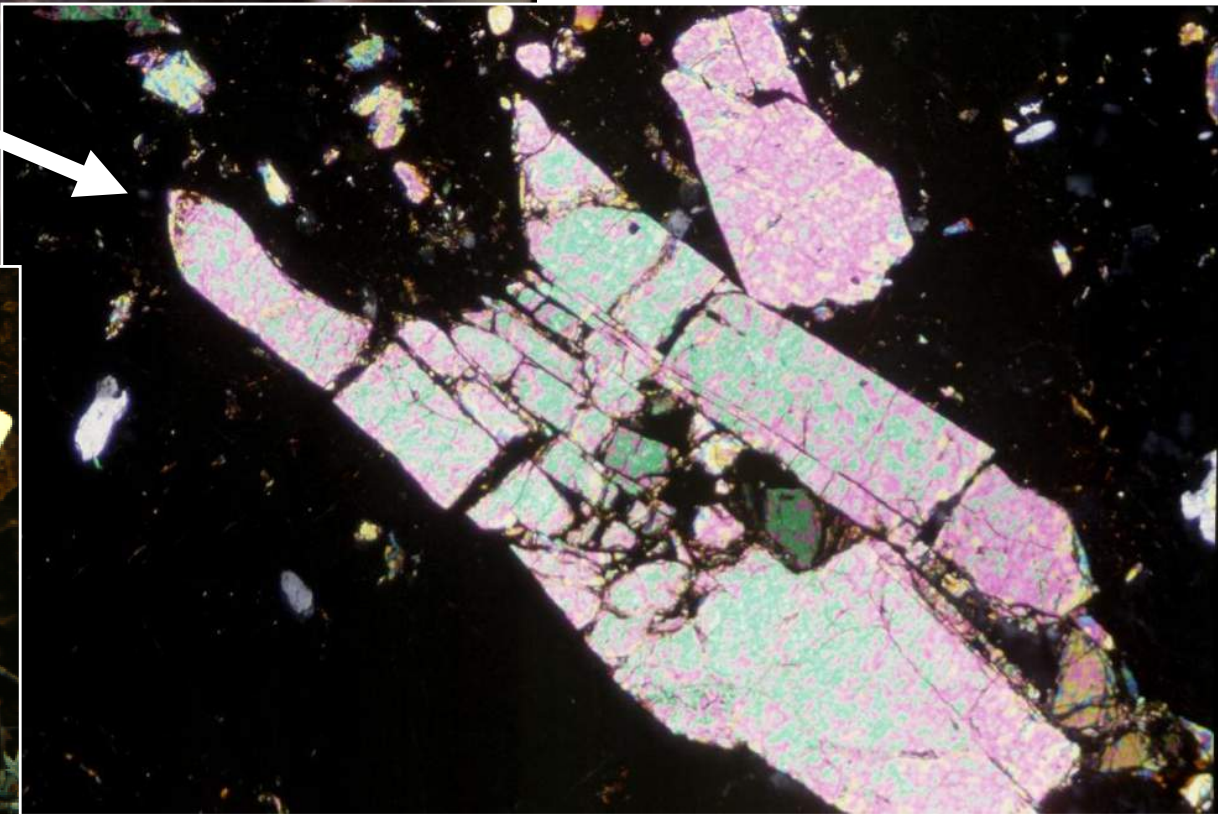


Gallinas Mountains,
New Mexico

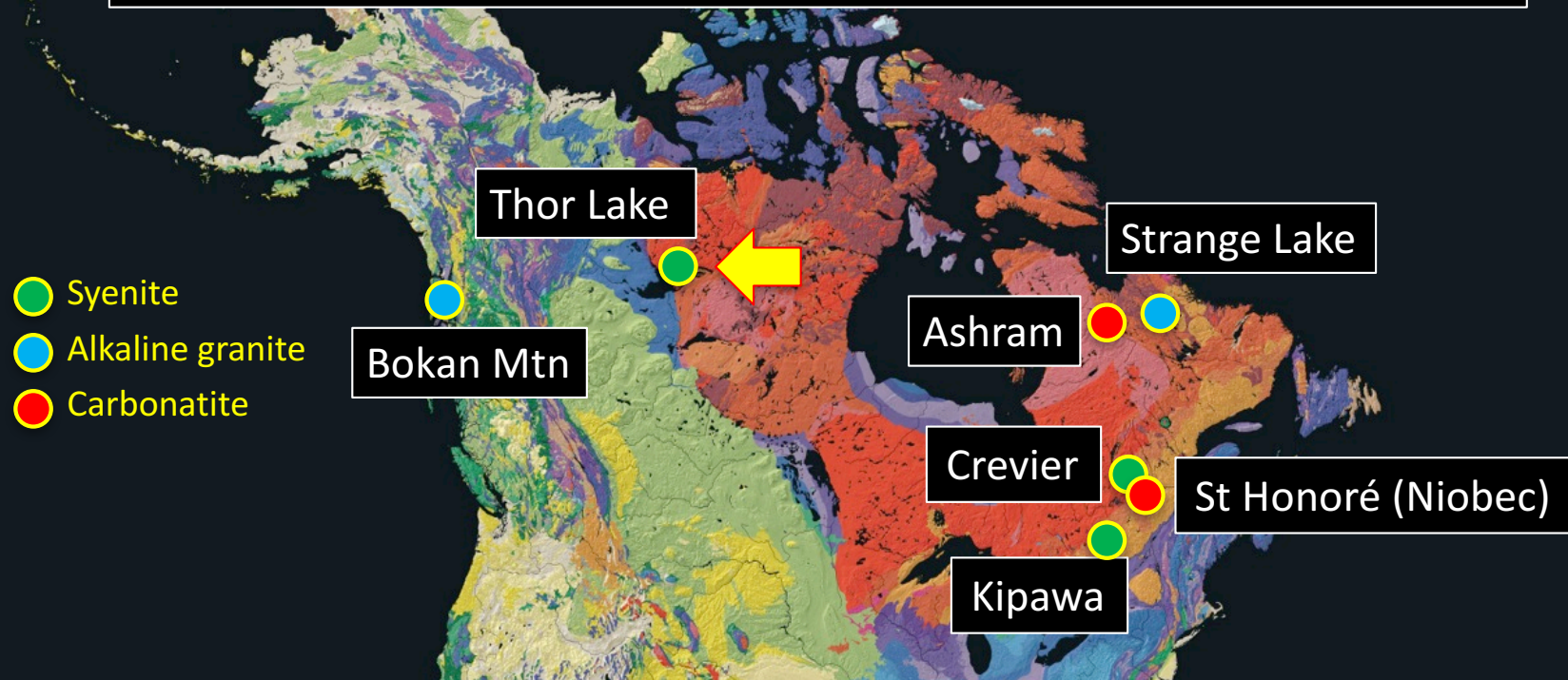
Hydrothermal veins
and breccias

abundant bastnäsite

bastnäsite



Alkaline-rock related rare metal deposits

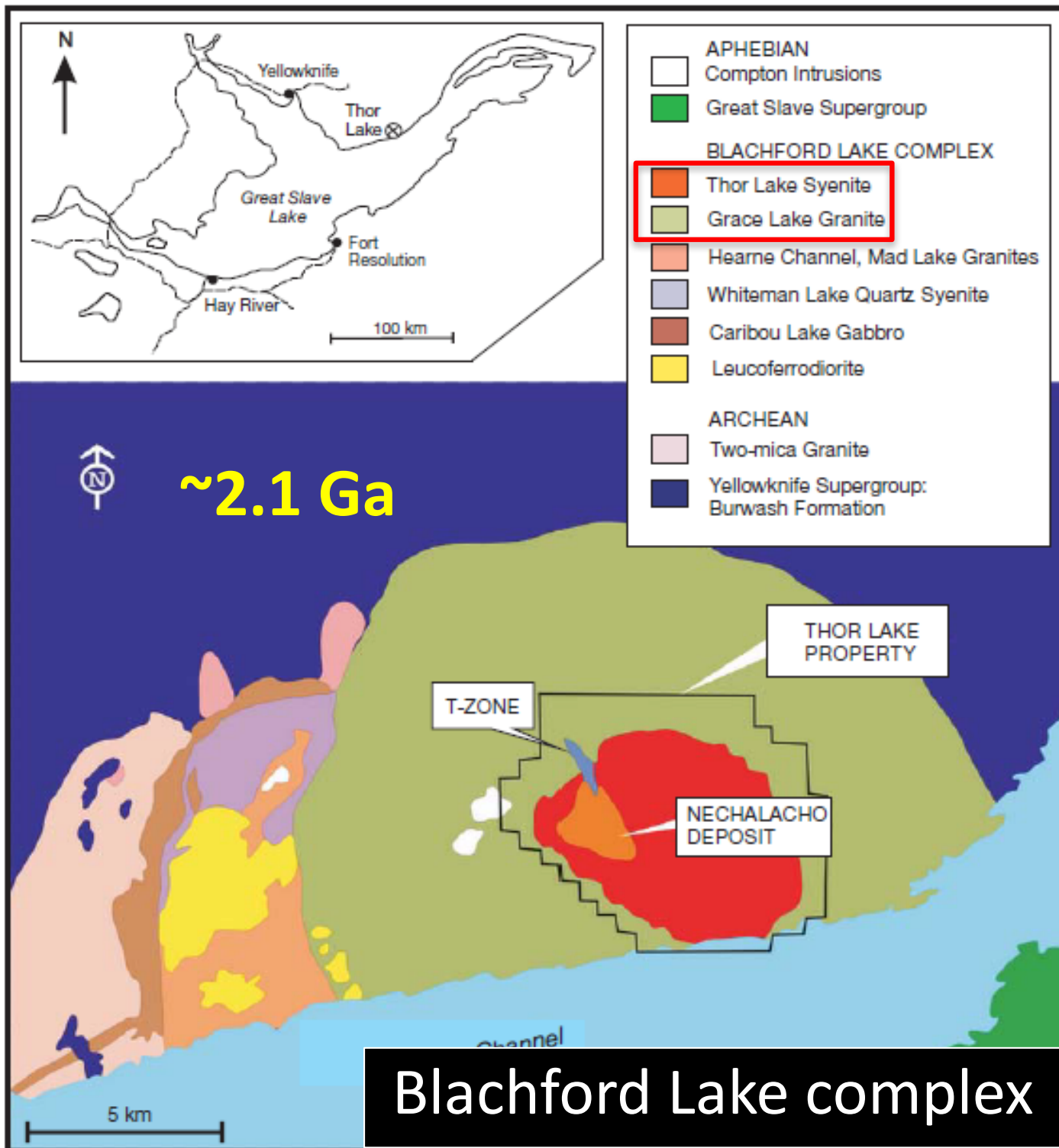


Questions: what roles do fluids play?

- Primary vs secondary concentration and enrichment?
- What types of fluids are capable of mobilizing significant rare metals?
- What is the evidence for this?
- Are rare metals present in the fluids?
- What concentrations?
- What controls enrichment?

Thor Lake Location





*modified after
Sheard et al., 2012;
Davidson, 1982*

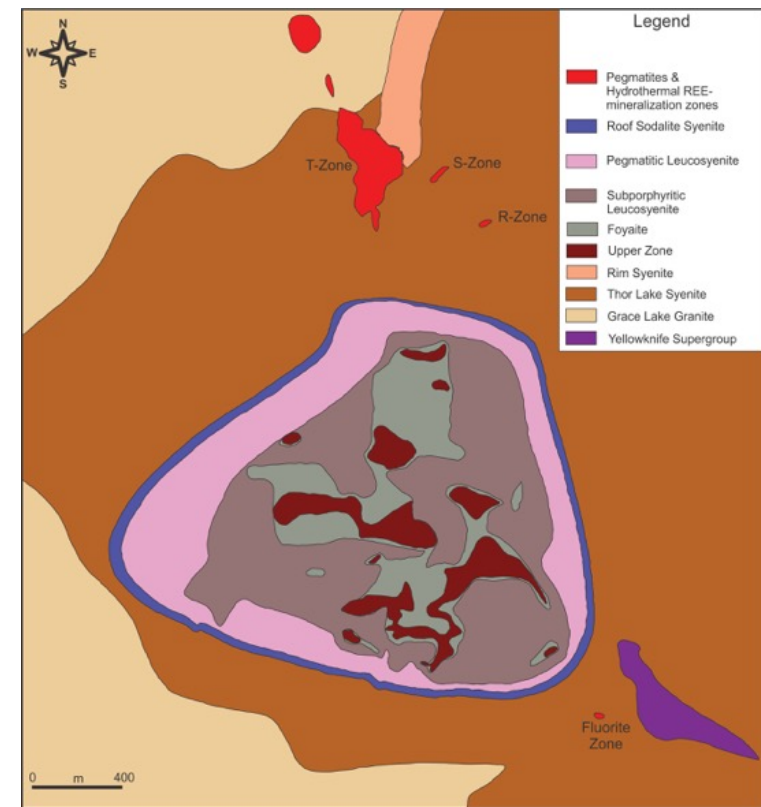
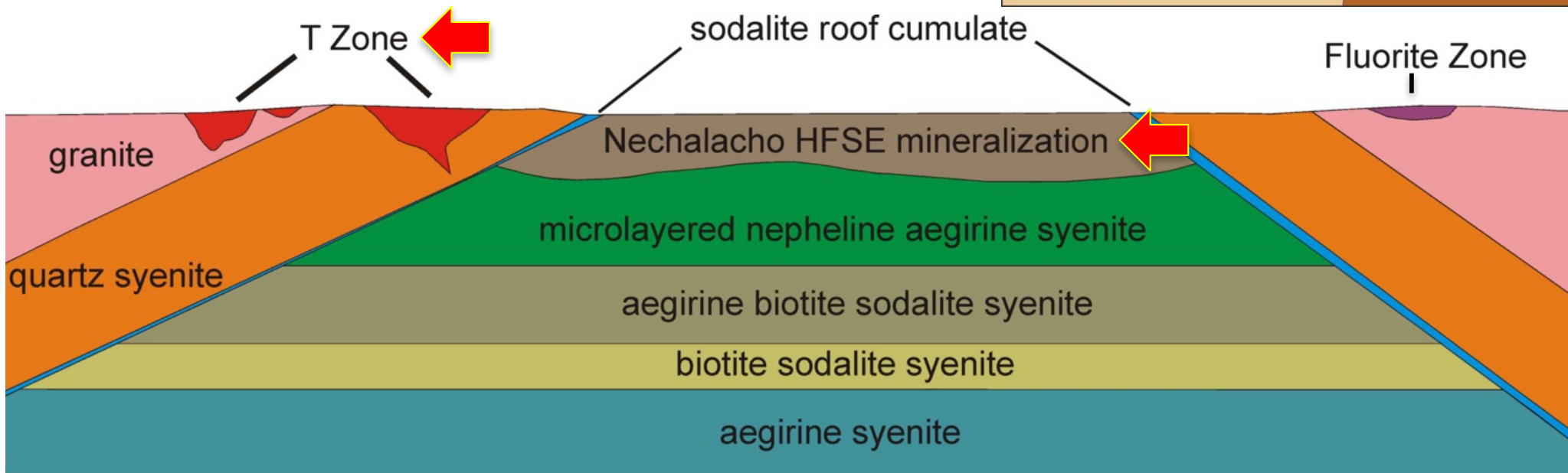
The Nechalacho Layered Series

Two Main Mineralized Zones

Nechalacho: REE, Zr, Nb

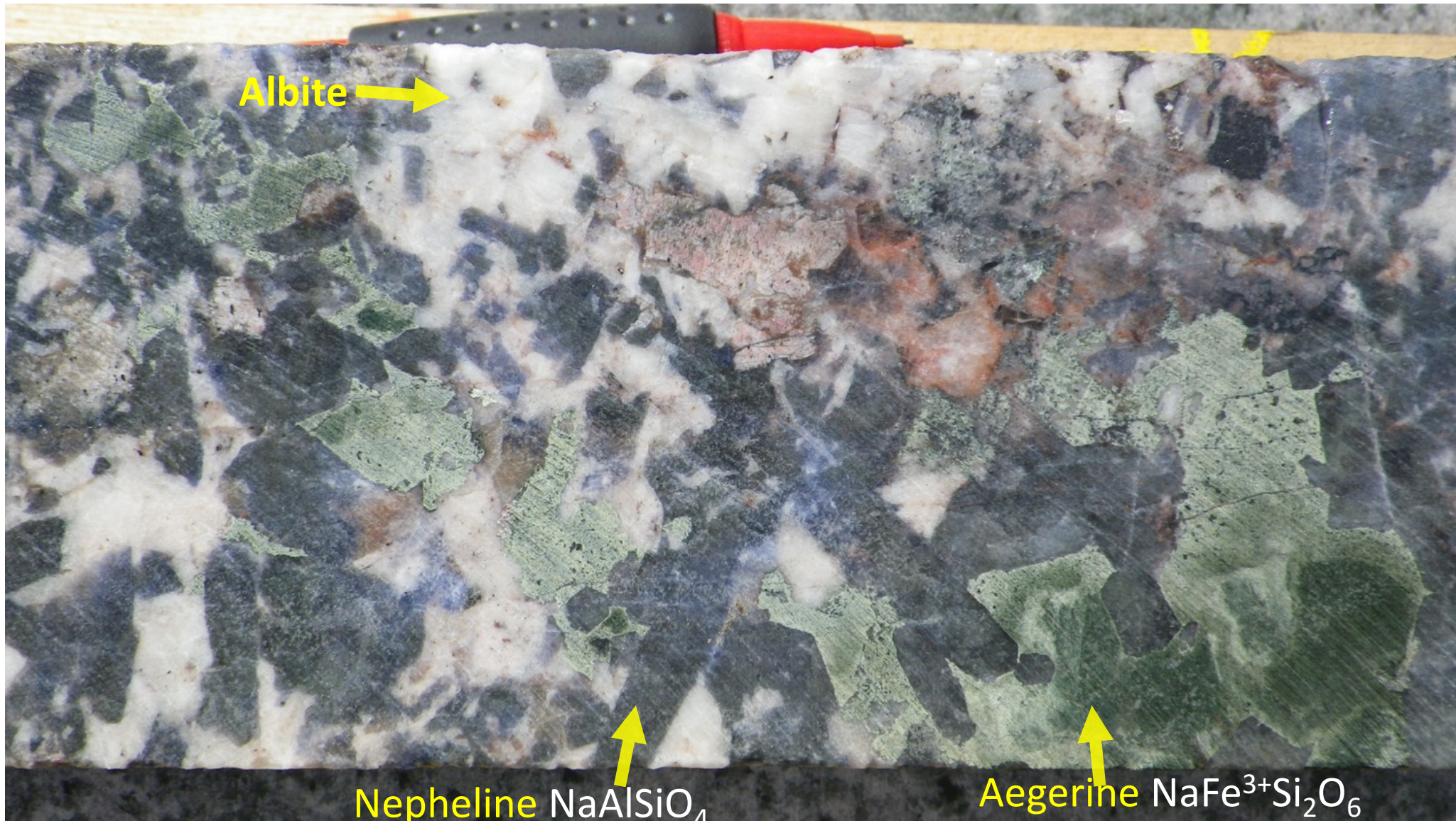
T Zone: REE, Zr, Nb, Li, Be

Schematic cross-section

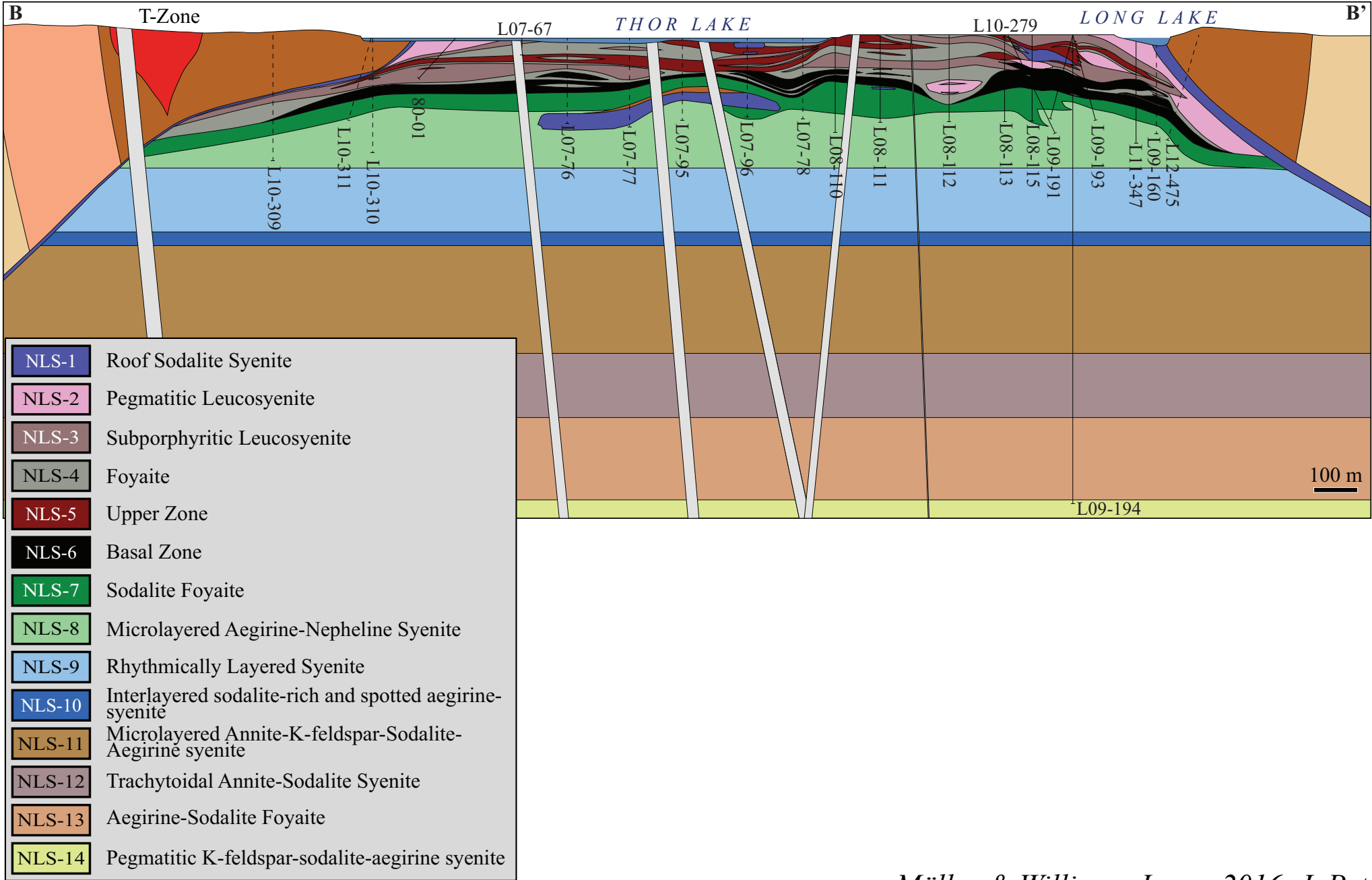


Modified after Möller and Williams-Jones (2012)

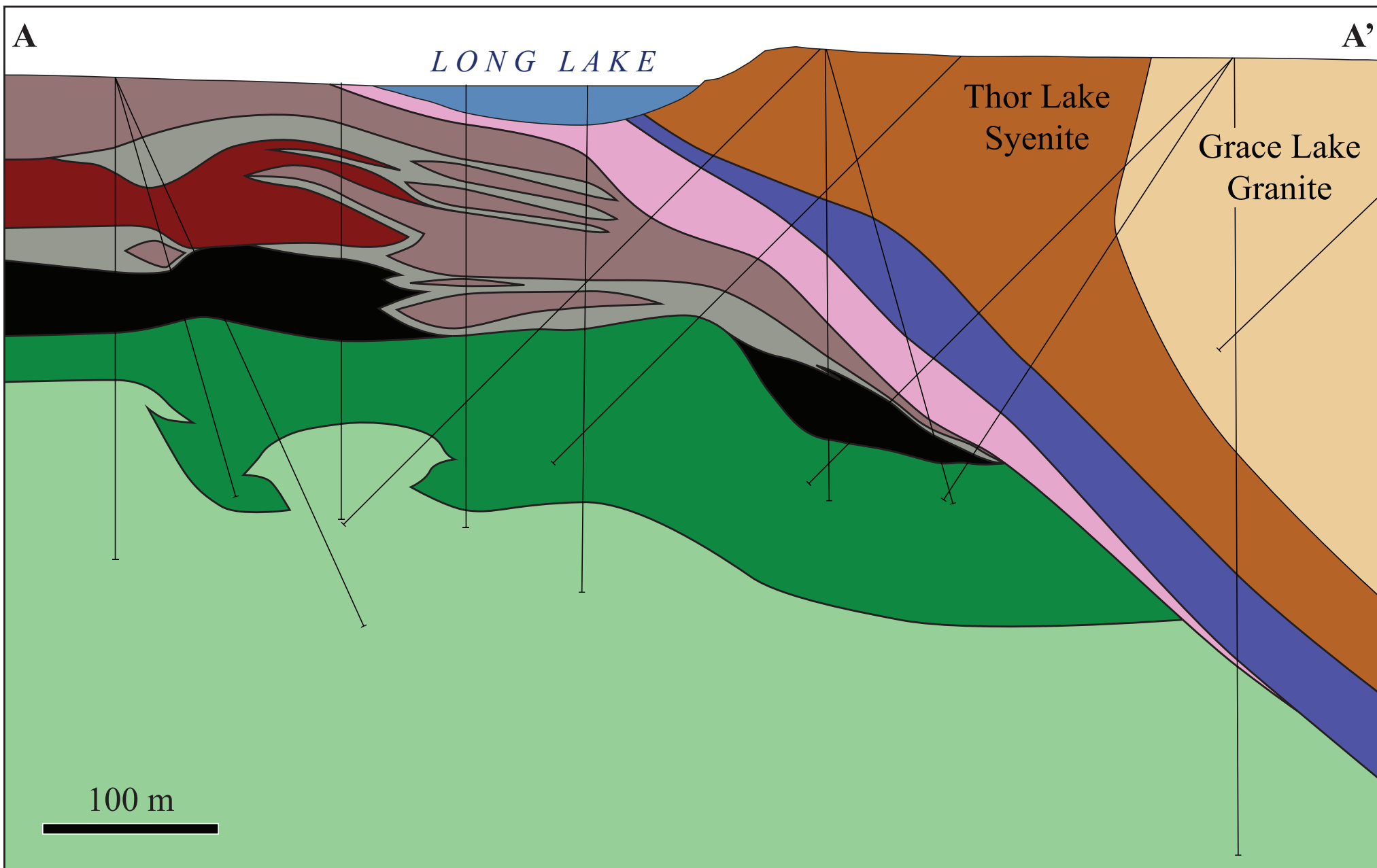
Nechalacho Layered Series: aegirine-nepheline-sodalite-biotite syenites



The Nechalacho Layered Suite



The Nechalacho Deposit



Nechalacho Deposit: Cumulates

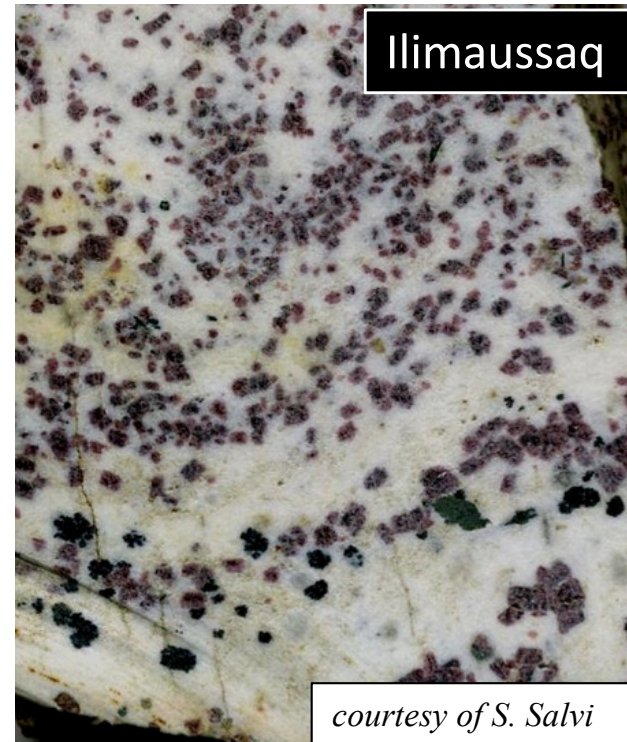
Thor Lake



Thor Lake



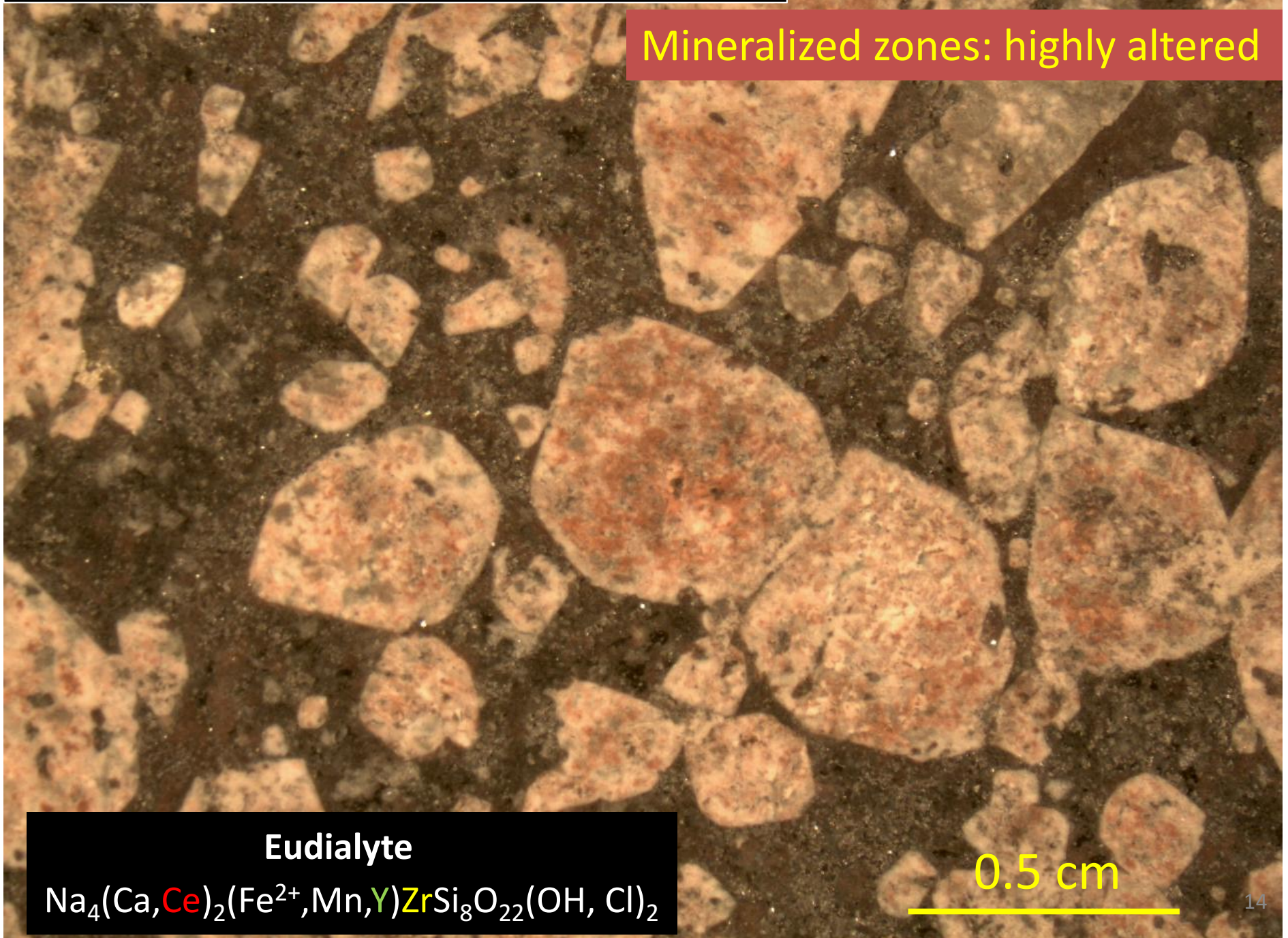
Ilimaussaq



courtesy of S. Salvi

Basal Zone Pseudomorphs

Mineralized zones: highly altered



Eudialyte



0.5 cm

Nechalacho: secondary rare-metal minerals

zircon ZrSiO_4

fergusonite LnNbO_4

columbite $(\text{Fe}, \text{Mn})(\text{Nb}, \text{Ta})_2\text{O}_6$

allanite $(\text{Ca}, \text{Na})_2\text{Ln}_3\text{Si}_6\text{O}_{18} \cdot 2\text{H}_2\text{O}$

bastnäsite $\text{Ln}(\text{CO}_3)\text{F}$

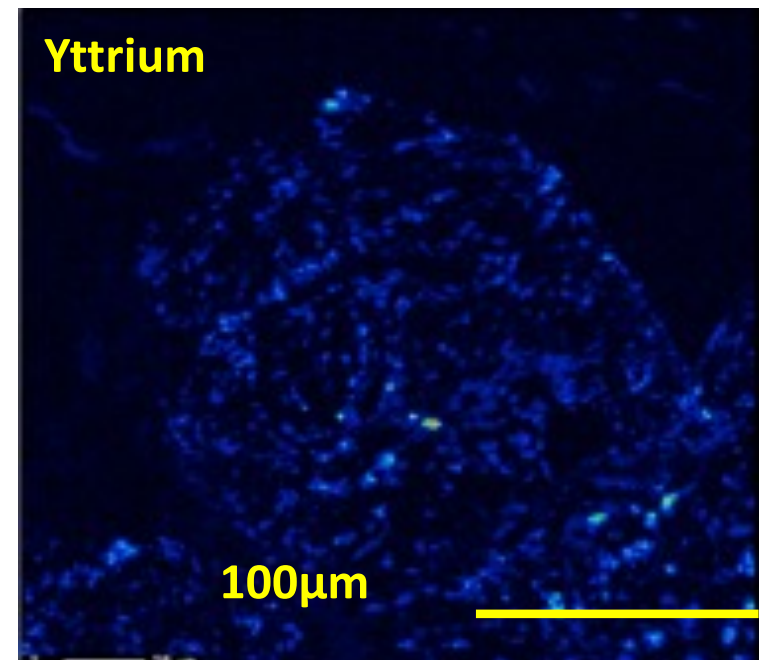
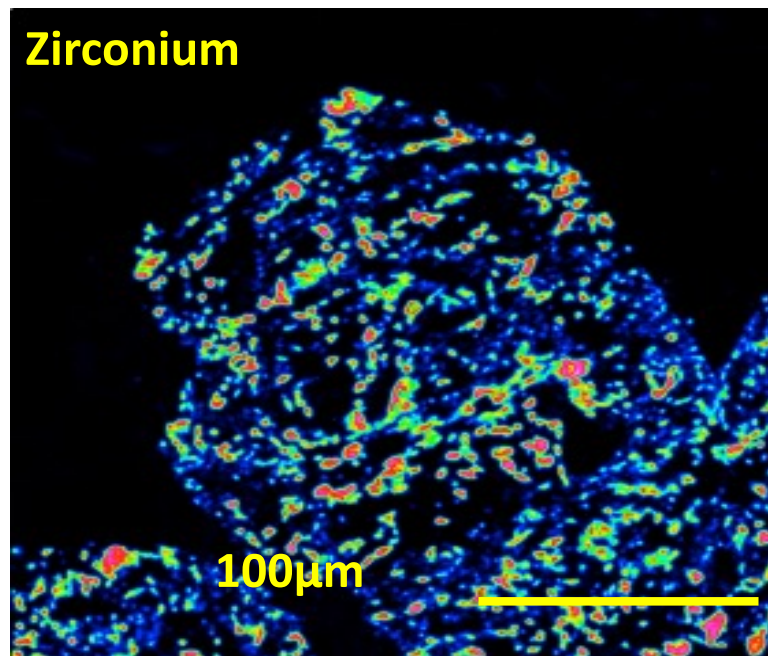
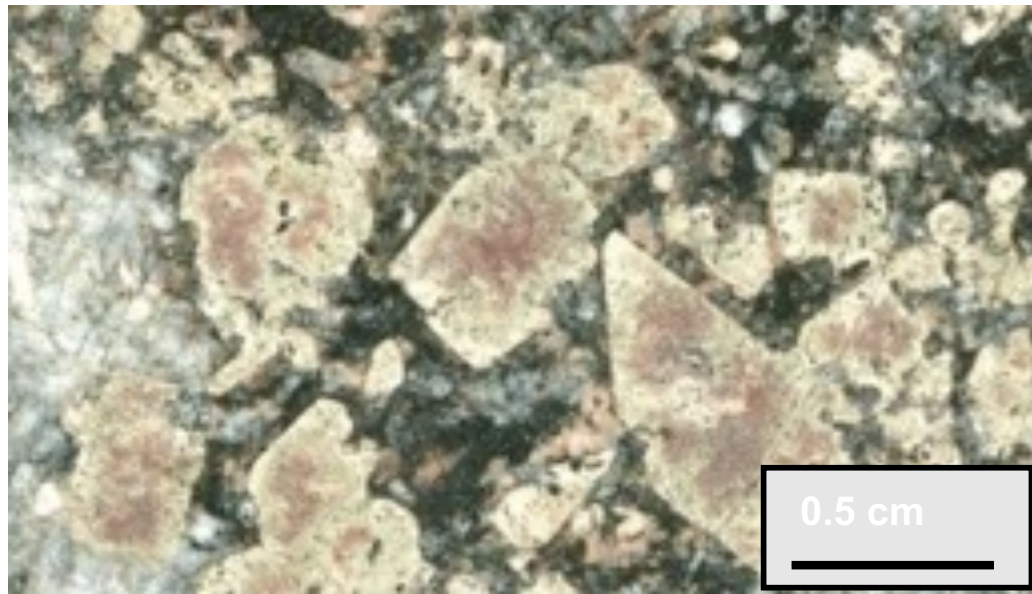
monazite LnPO_4 (LREE-enriched)

xenotime $(\text{Y}, \text{Ln})\text{PO}_4$ (HREE-enriched)

Ln = lanthanide

eudialyte pseudomorphs

Courtesy of E. Sheard

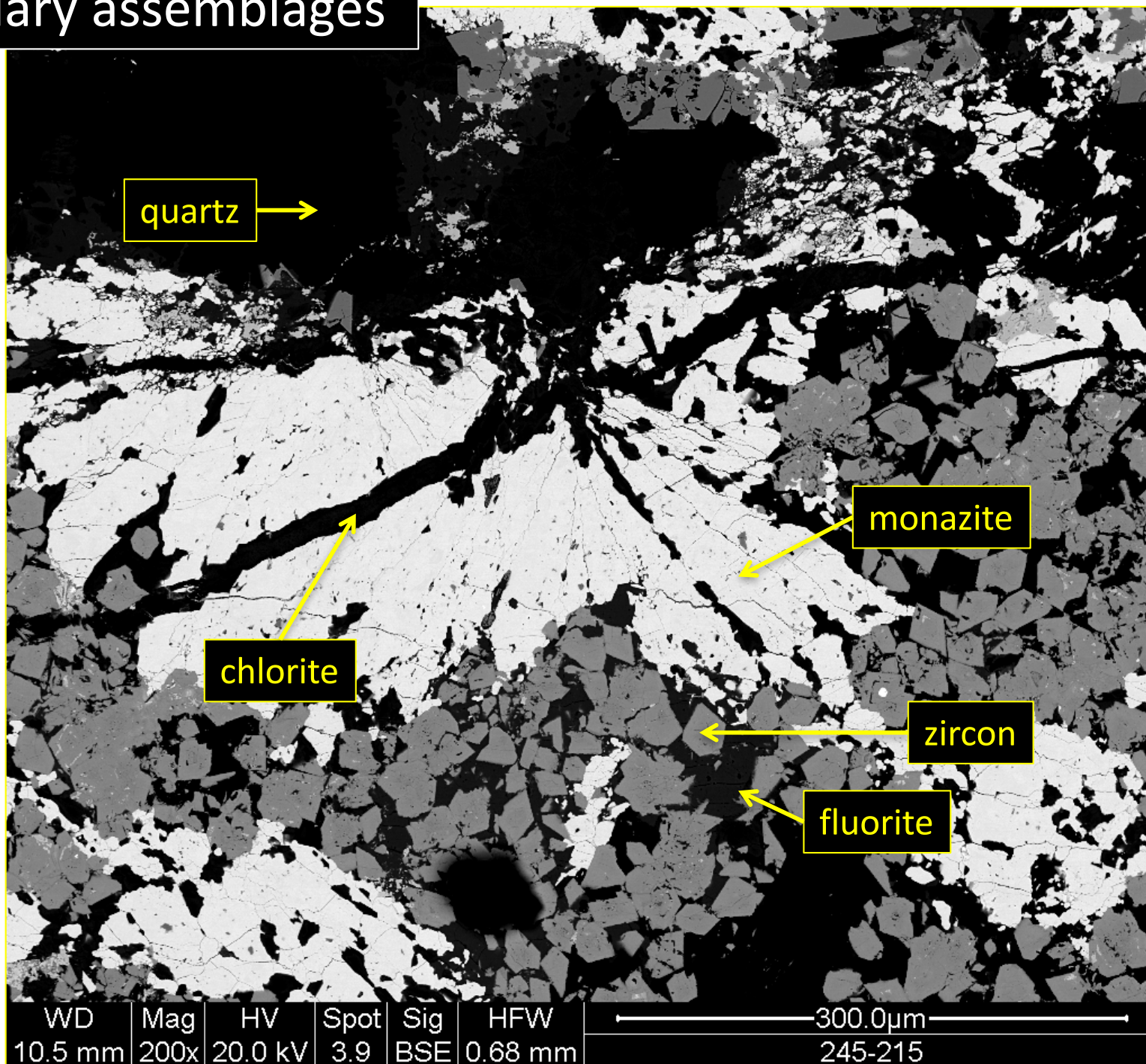


Heavily altered Basal Zone

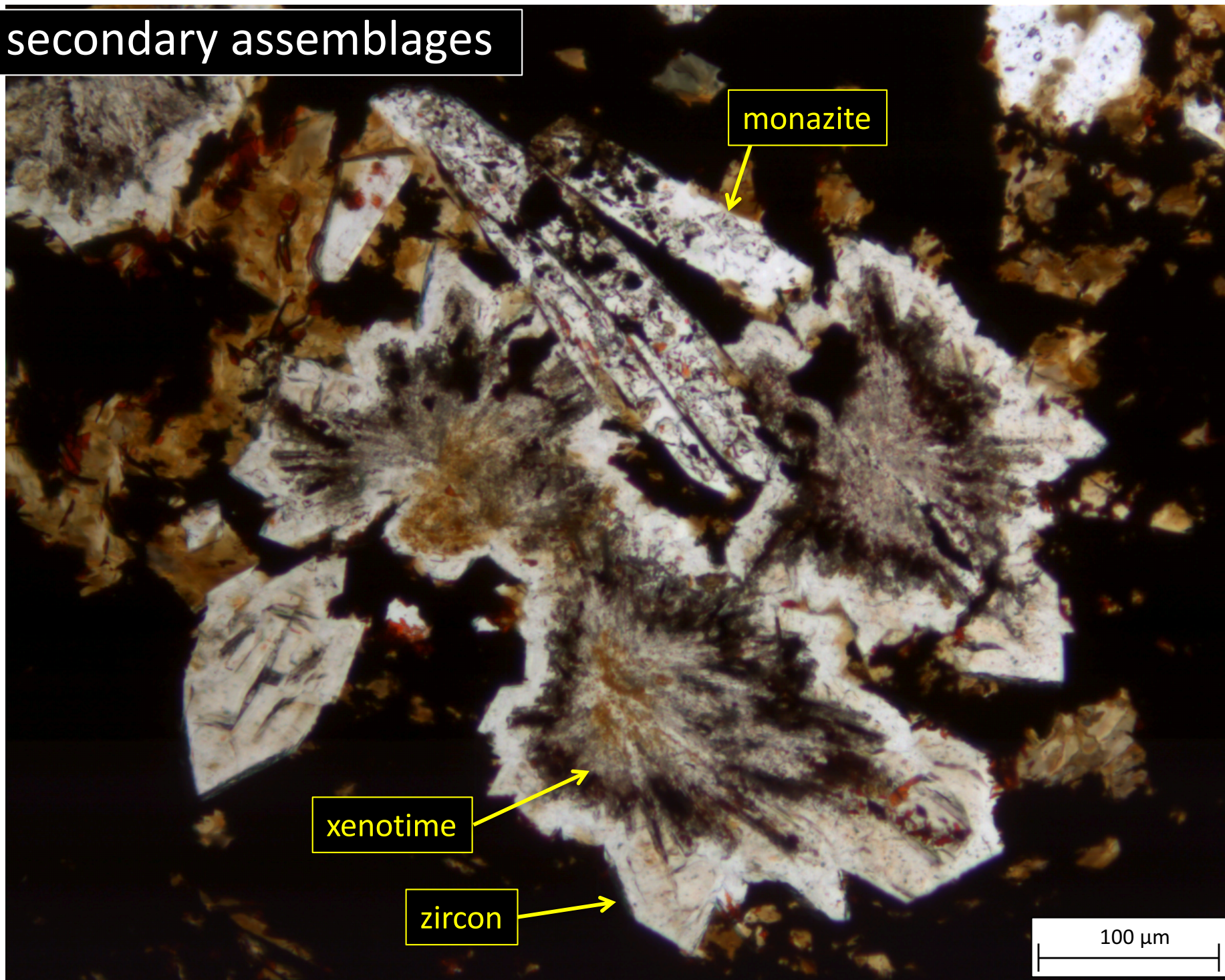
L10-245 210.25m-210.34m



secondary assemblages



secondary assemblages



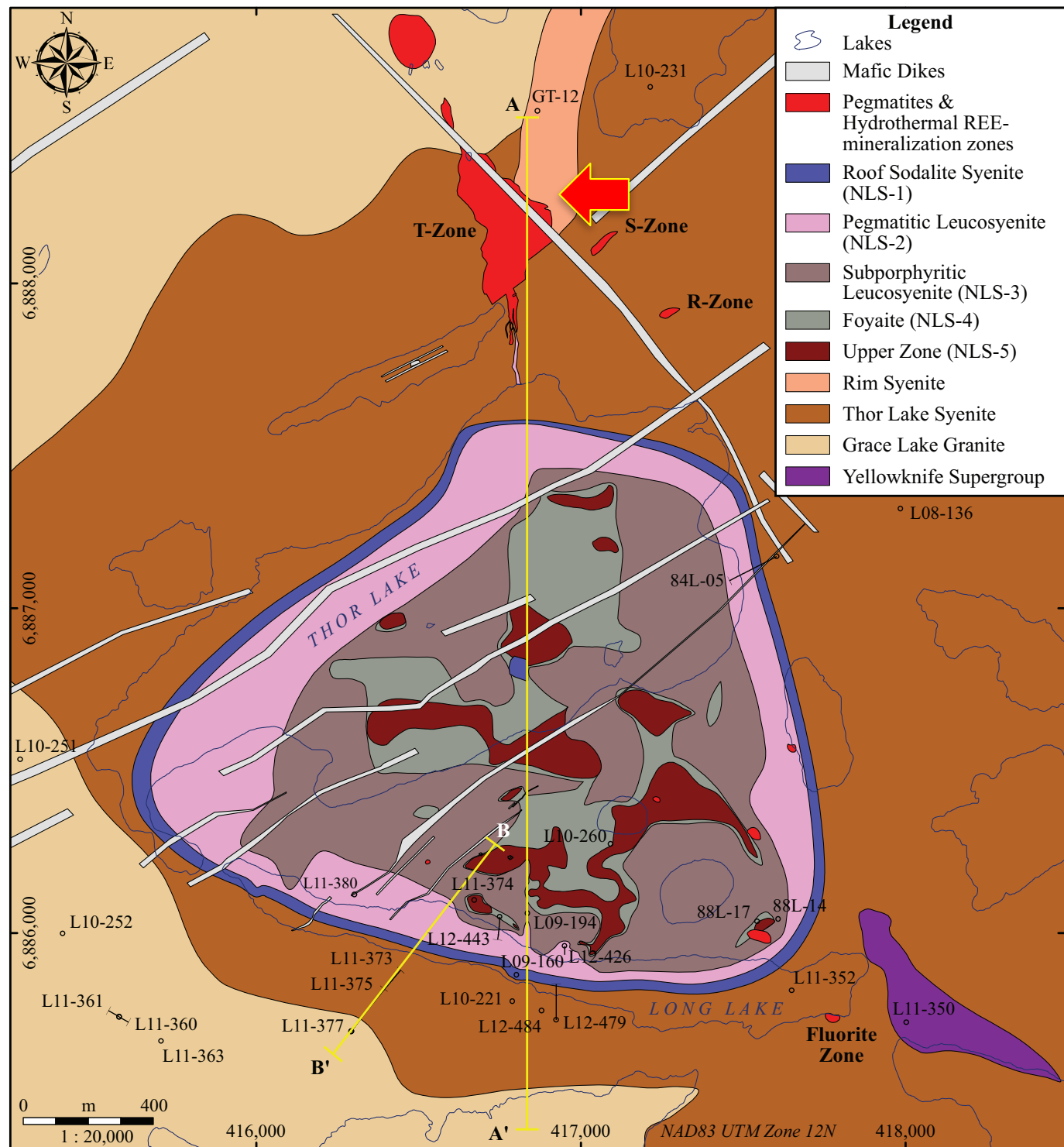
monazite

xenotime

zircon

100 μm

T Zone

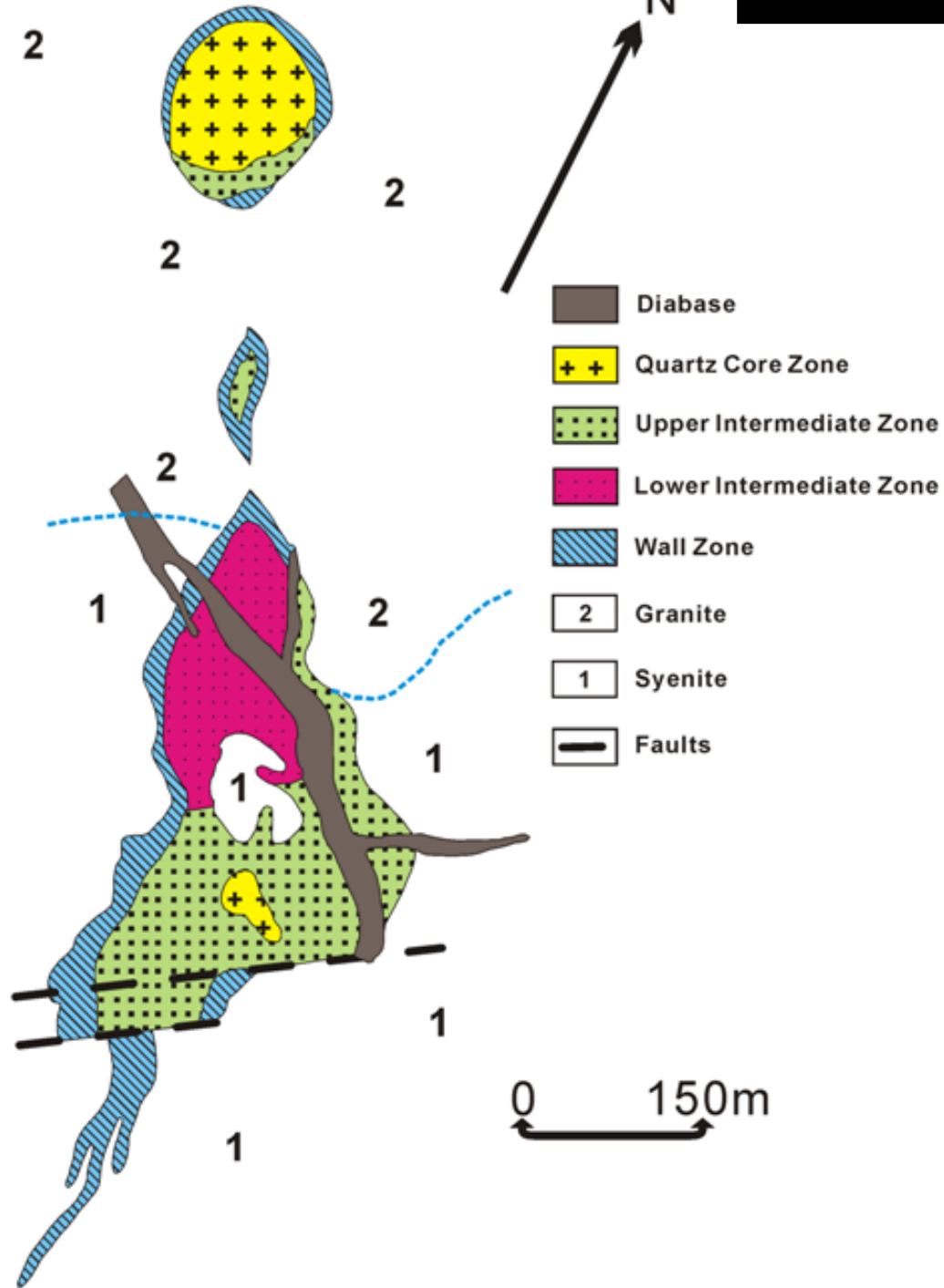


Möller & Williams-Jones, 2016, J. Pet.

T Zone Geology

b

2



0 150m

Pegmatitic Textures

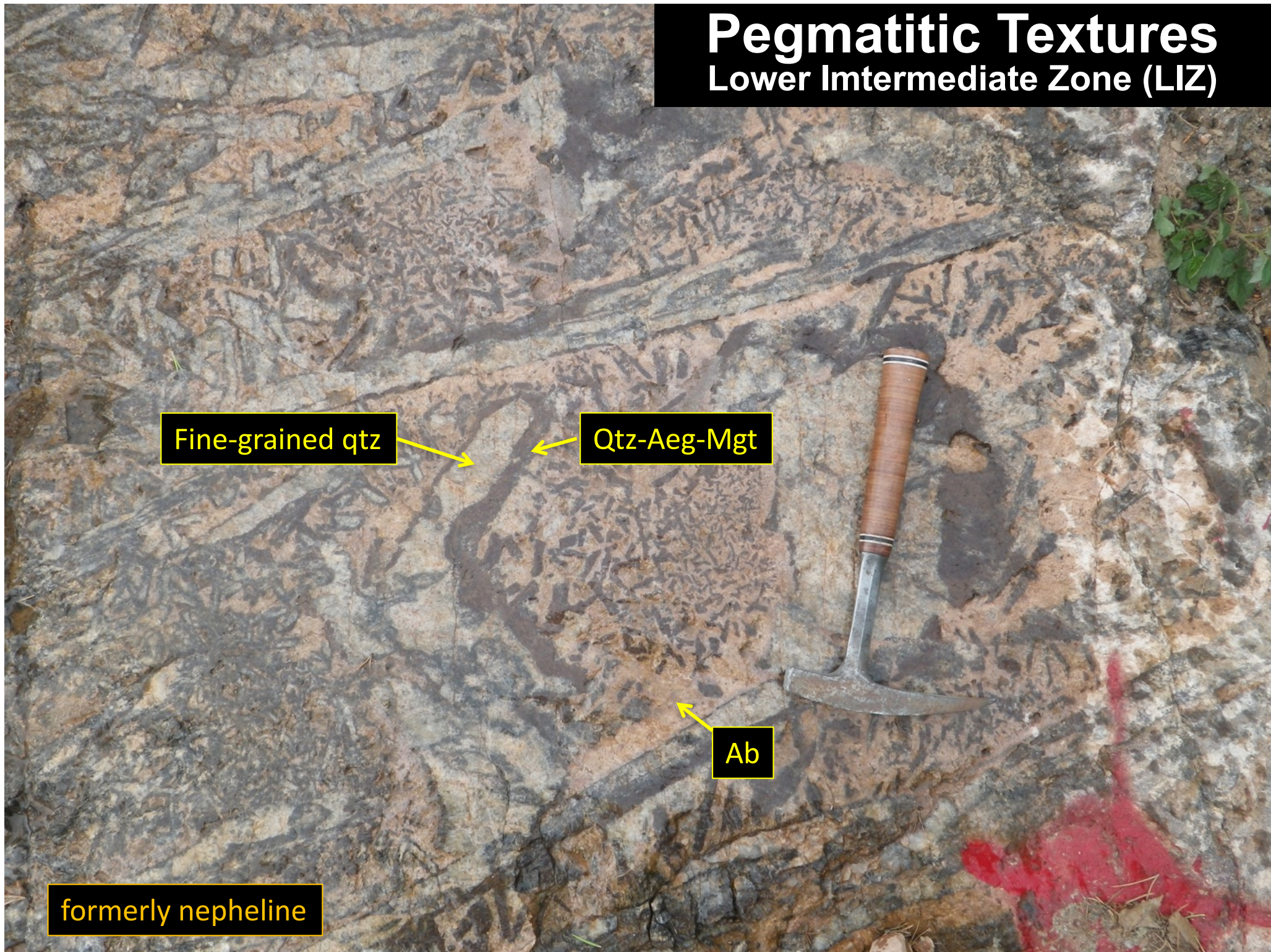
Lower Intermediate Zone (LIZ)

Fine-grained qtz

Qtz-Aeg-Mgt

Ab

formerly nepheline



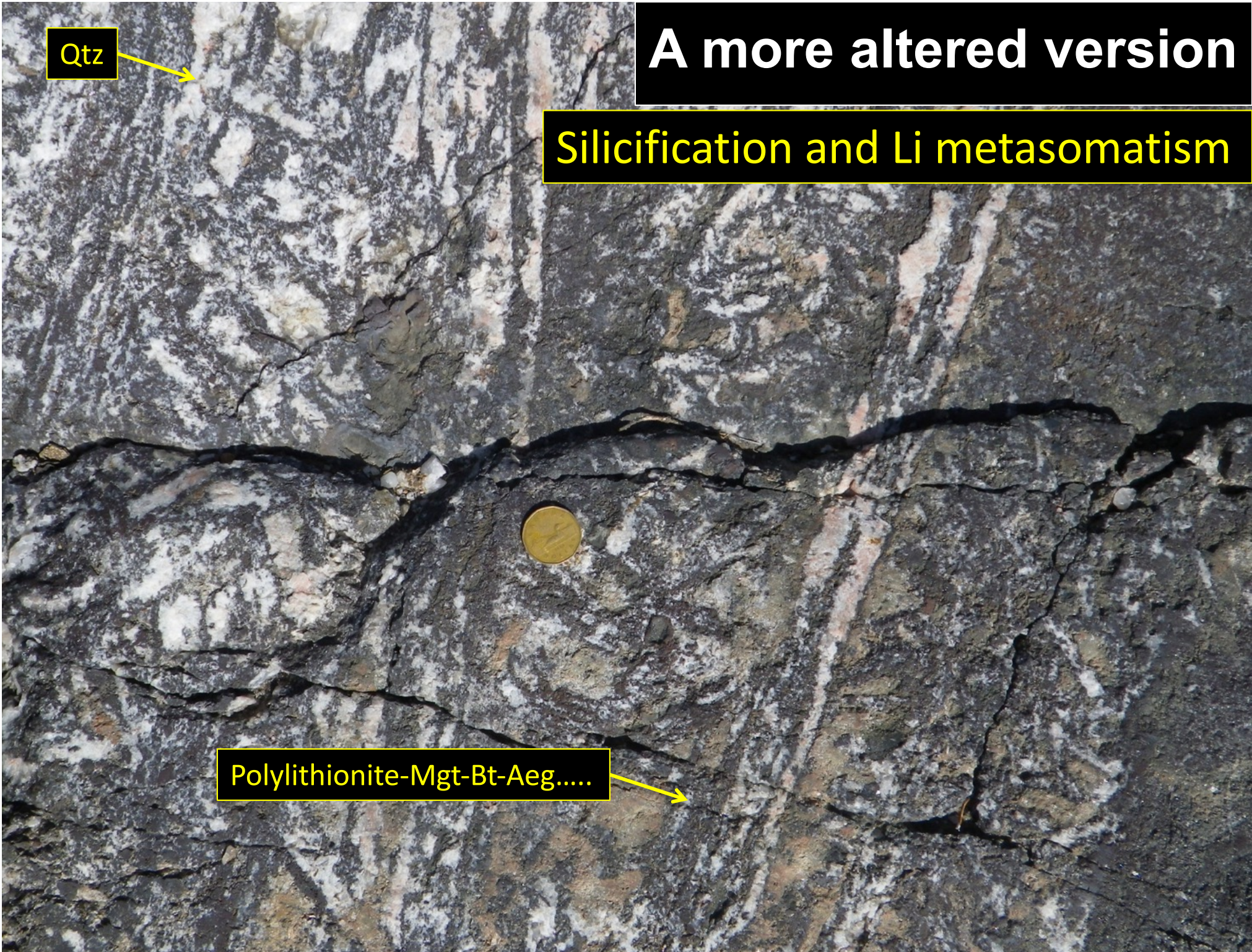
A more altered version

Silicification and Li metasomatism

Qtz



Polylithionite-Mgt-Bt-Aeg....



LIZ

Ply = Polylithionite: $\text{KLi}_2\text{Al}(\text{Si}_4\text{O}_{10})(\text{F},\text{OH})_2$

Ply

UIZ



d

T Zone rare-metal minerals

zircon ZrSiO_4

columbite $(\text{Fe}, \text{Mn})(\text{Nb}, \text{Ta})_2\text{O}_6$

bastnäsite $\text{Ln}(\text{CO}_3)\text{F}$

monazite LnPO_4

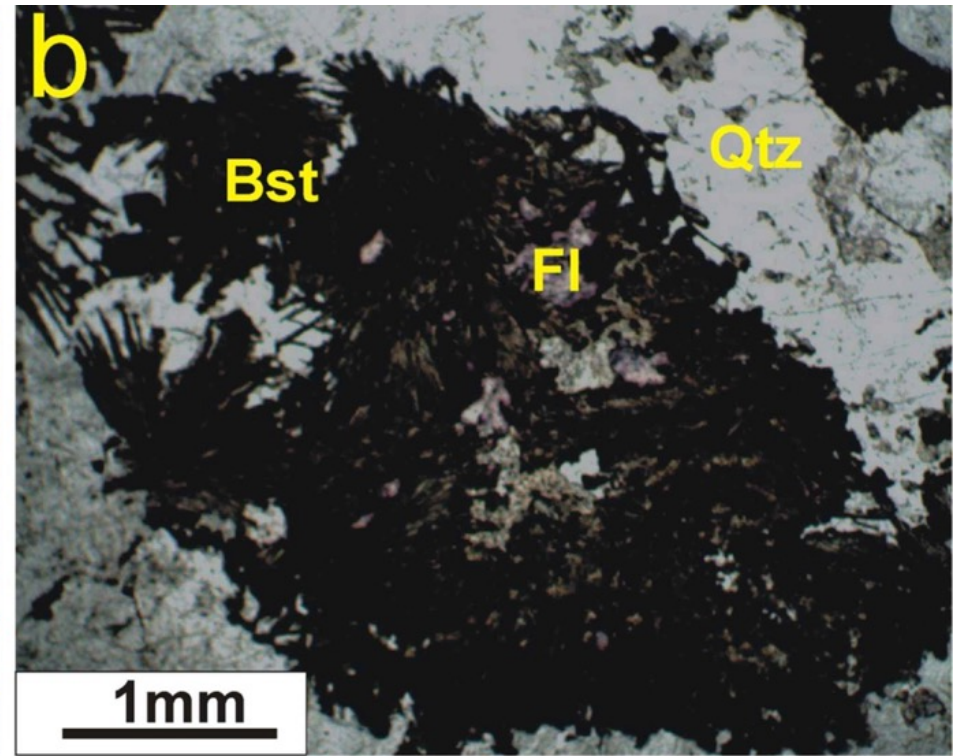
xenotime $(\text{Y}, \text{Ln})\text{PO}_4$

phenakite Be_2SiO_4

polyolithionite $\text{KLi}_2\text{AlSi}_4\text{O}_{10}(\text{F}, \text{OH})_2$

How do the rare-metal minerals occur?
are they primary or secondary?

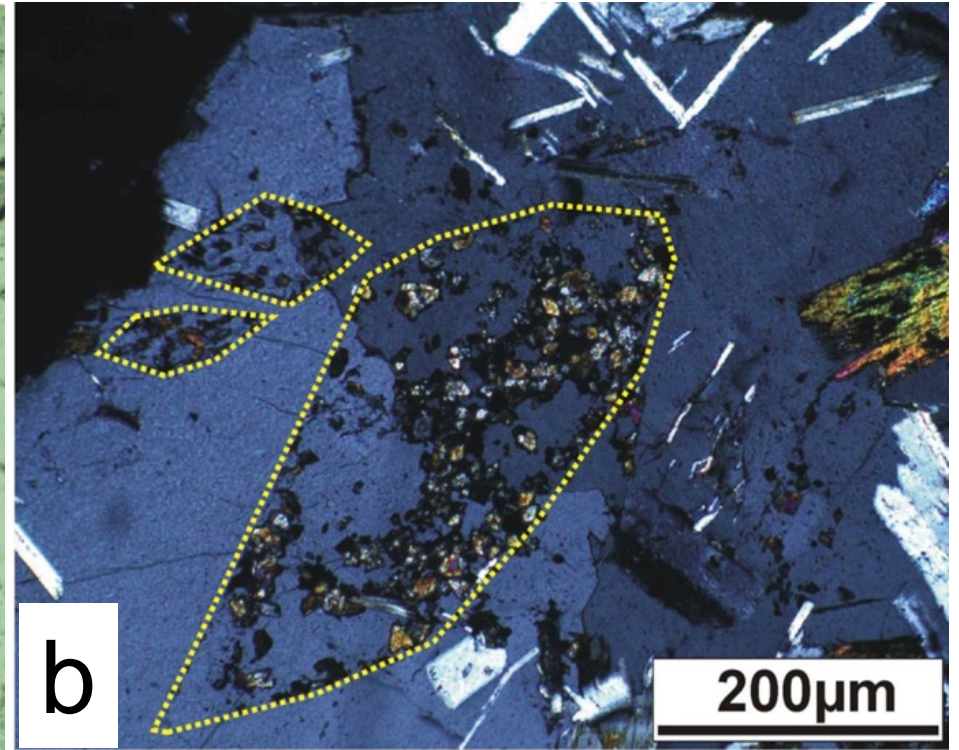
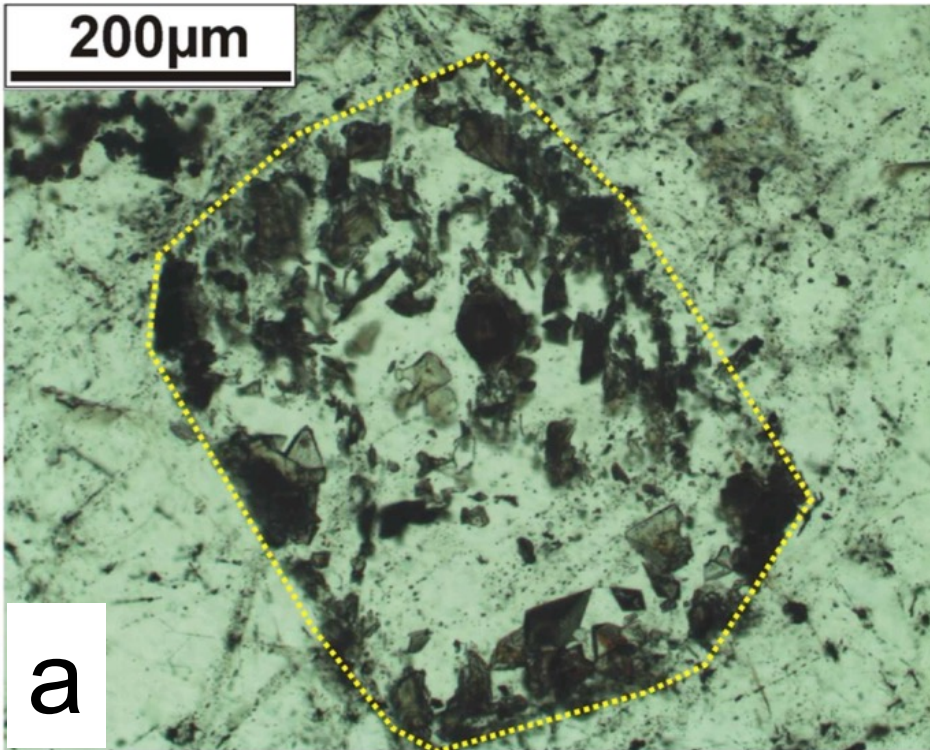
REE-minerals in pseudomorphs



Bst = bastnäsite (REECO_3F), LREE-rich mineral

Also: monazite, xenotime

Zircon in pseudomorphs

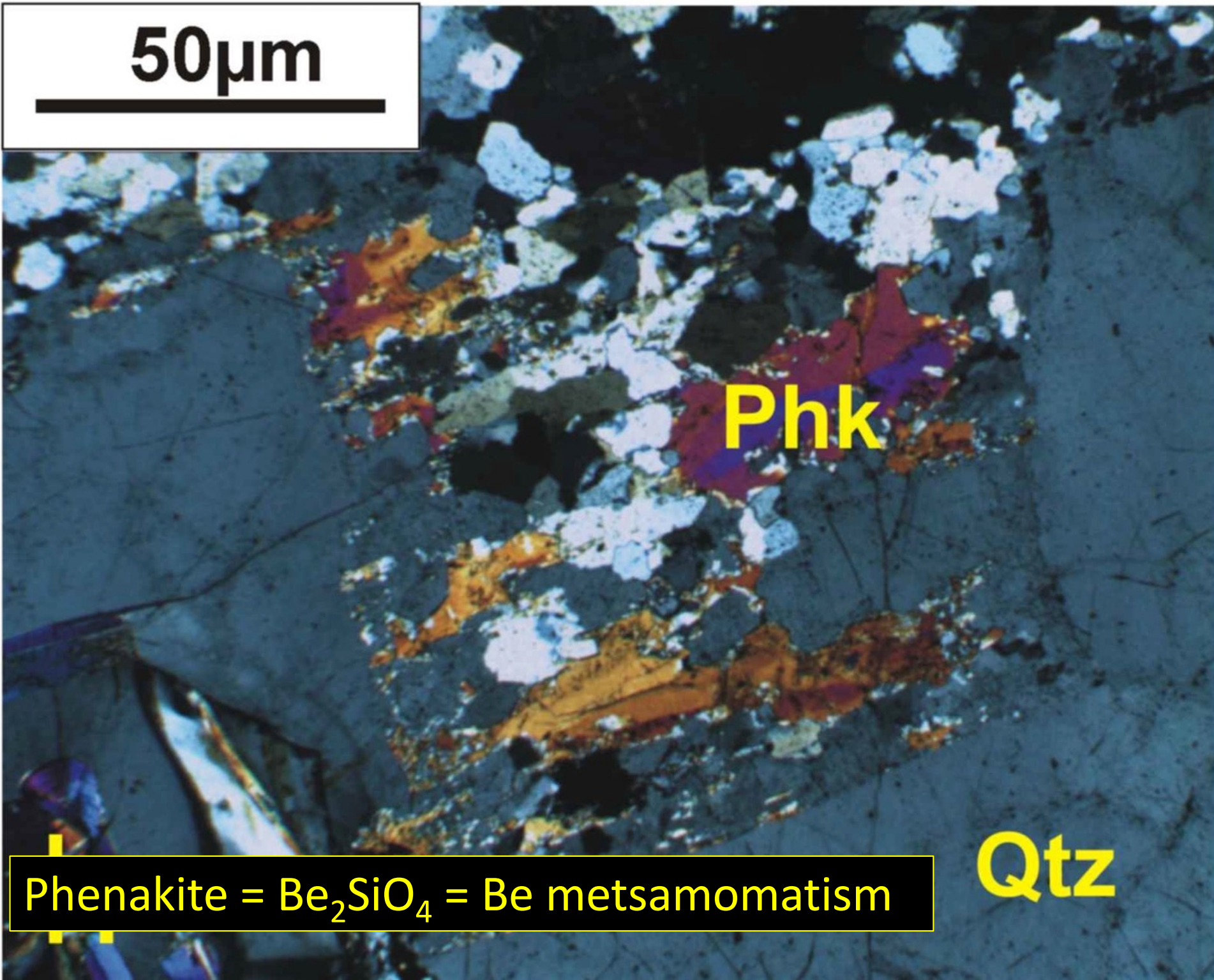


50μm

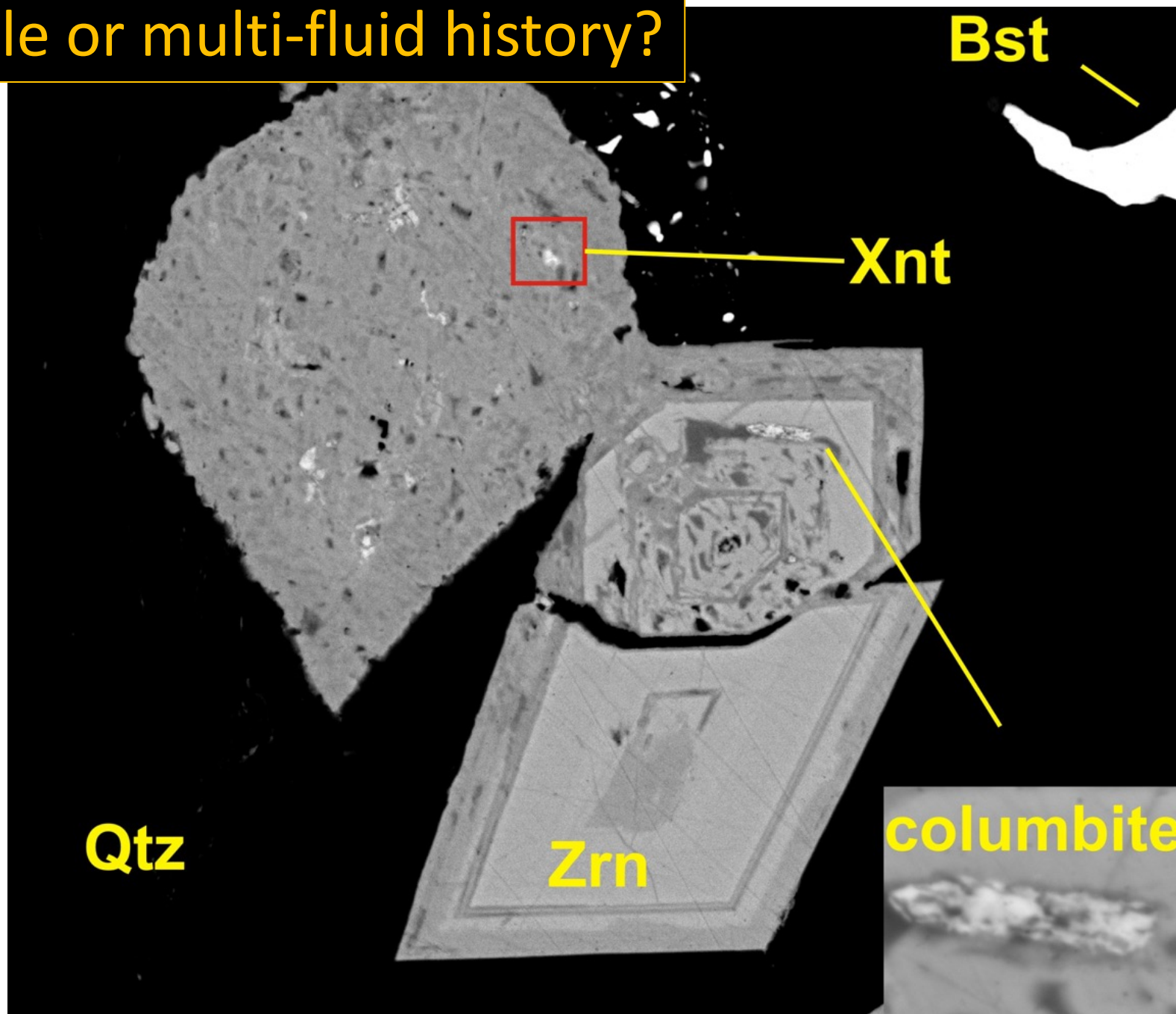
Phk

Phenakite = Be_2SiO_4 = Be metamatism

Qtz



Simple or multi-fluid history?



multiple generations of zircon

HFW
0.63 μm

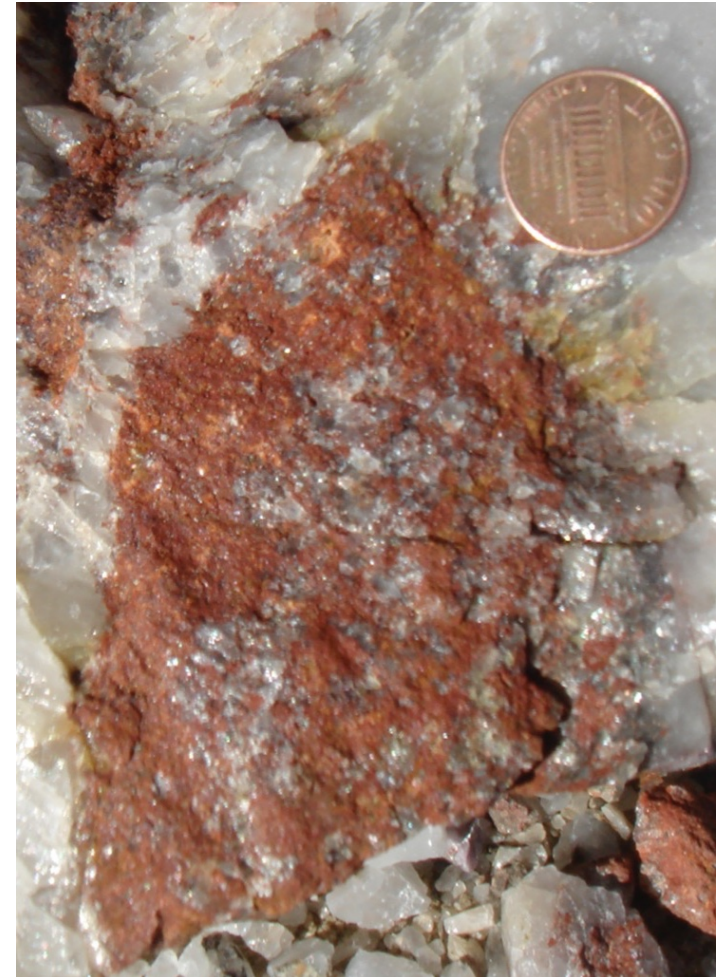
20.0 μm
X28-B7-F1

What was the character of the fluids and did they contain rare-metals?

Does the fluid record bear-out the mineralogical complexity?

Approach = fluid inclusions

How can we tie the fluid inclusions to the rare-metal minerals?



fluid inclusions restricted to pseudomorphs

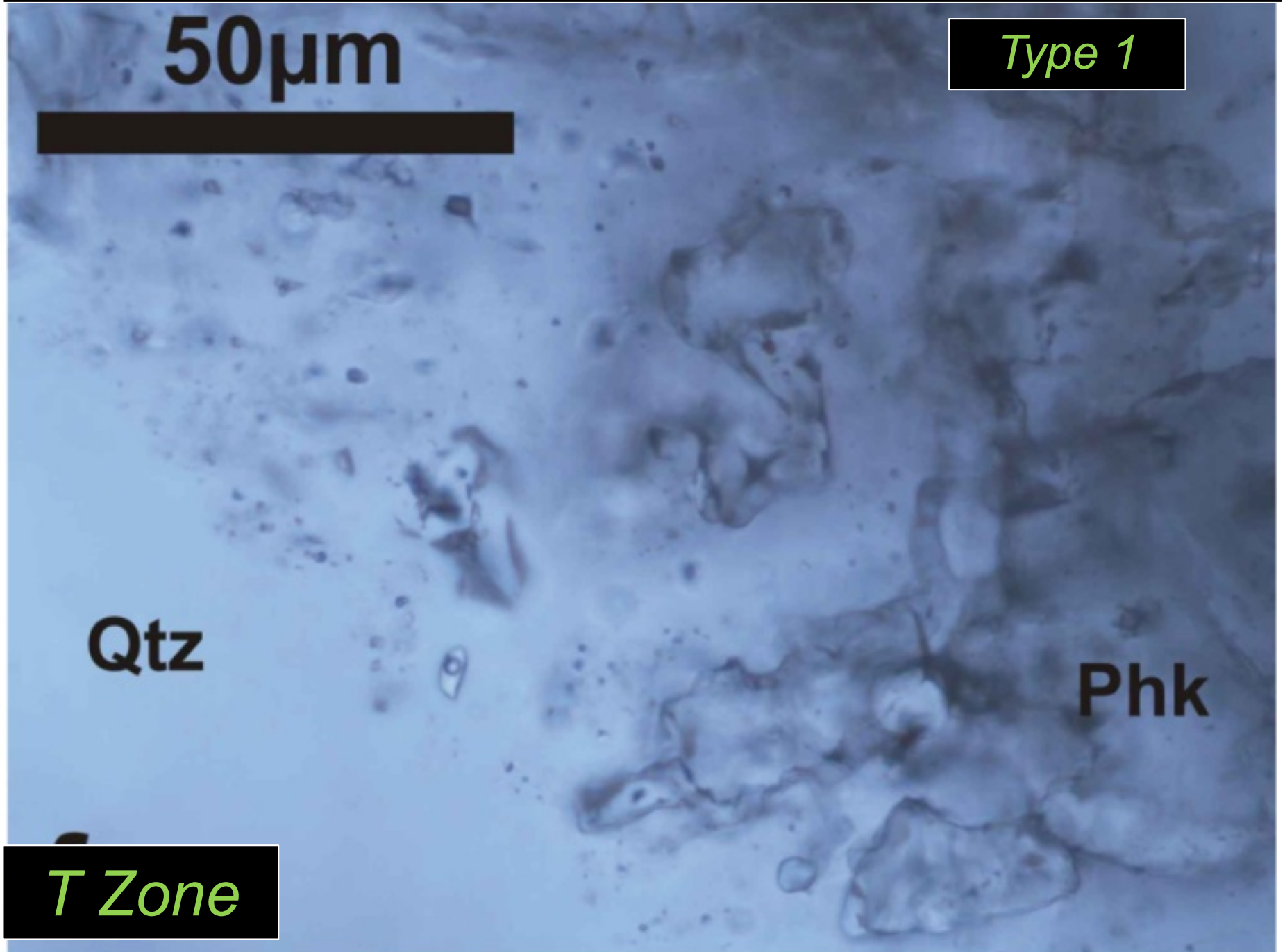
50 μ m

Type 1

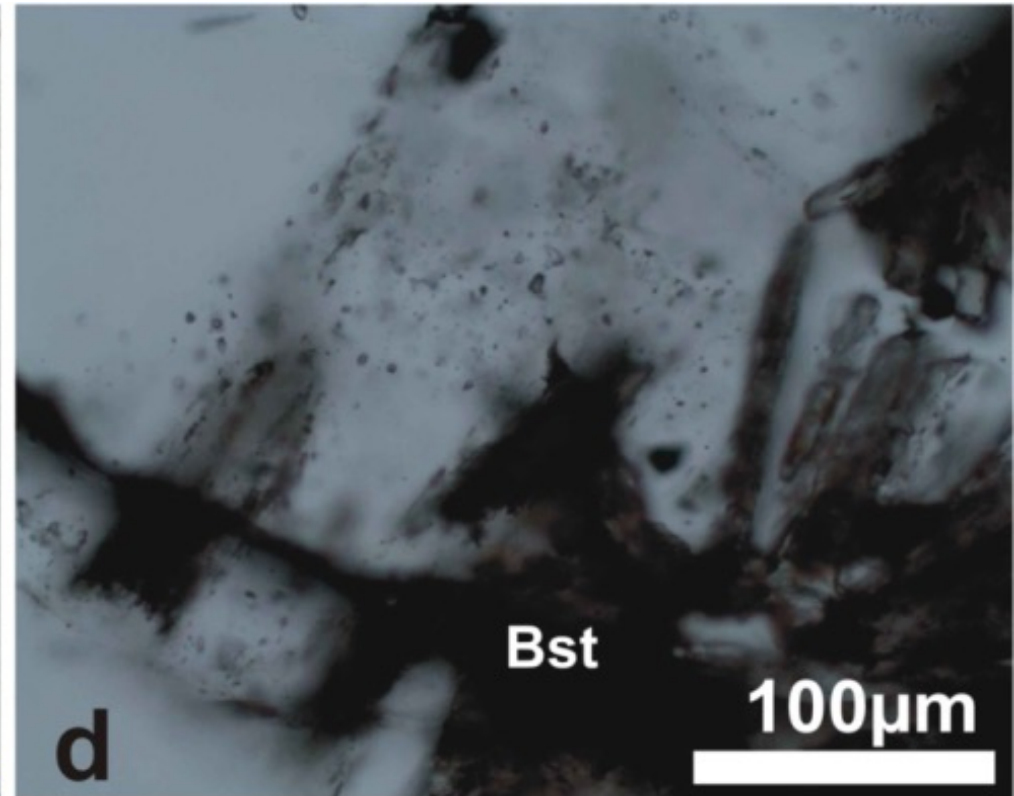
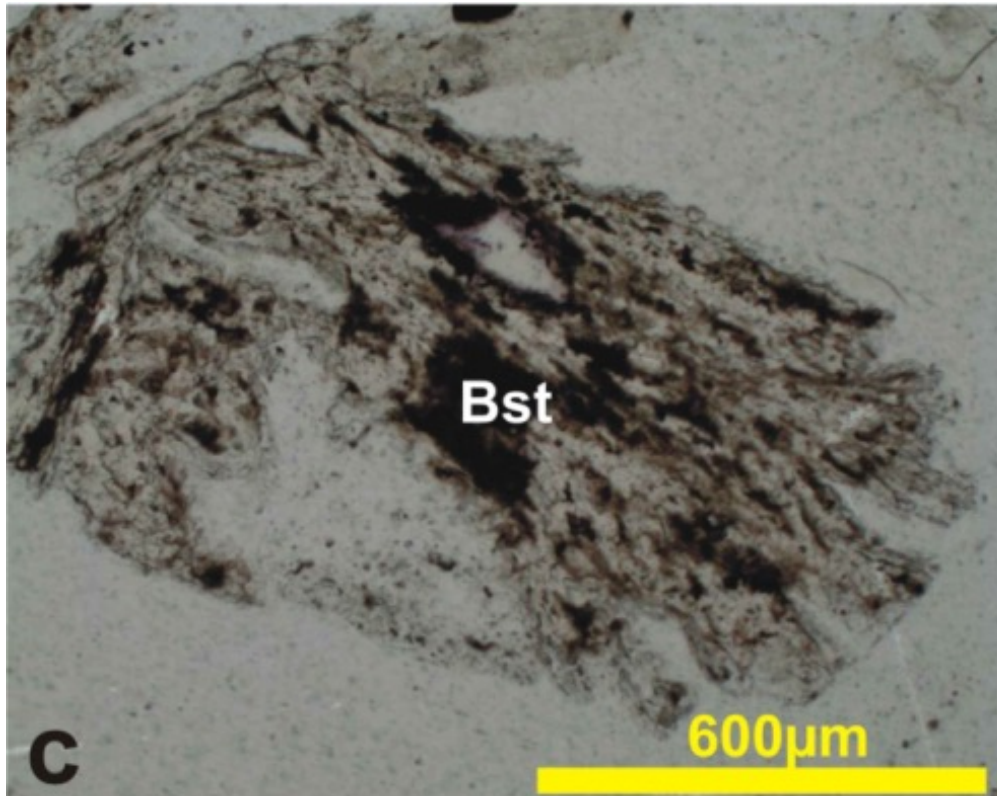
Qtz

Phk

T Zone

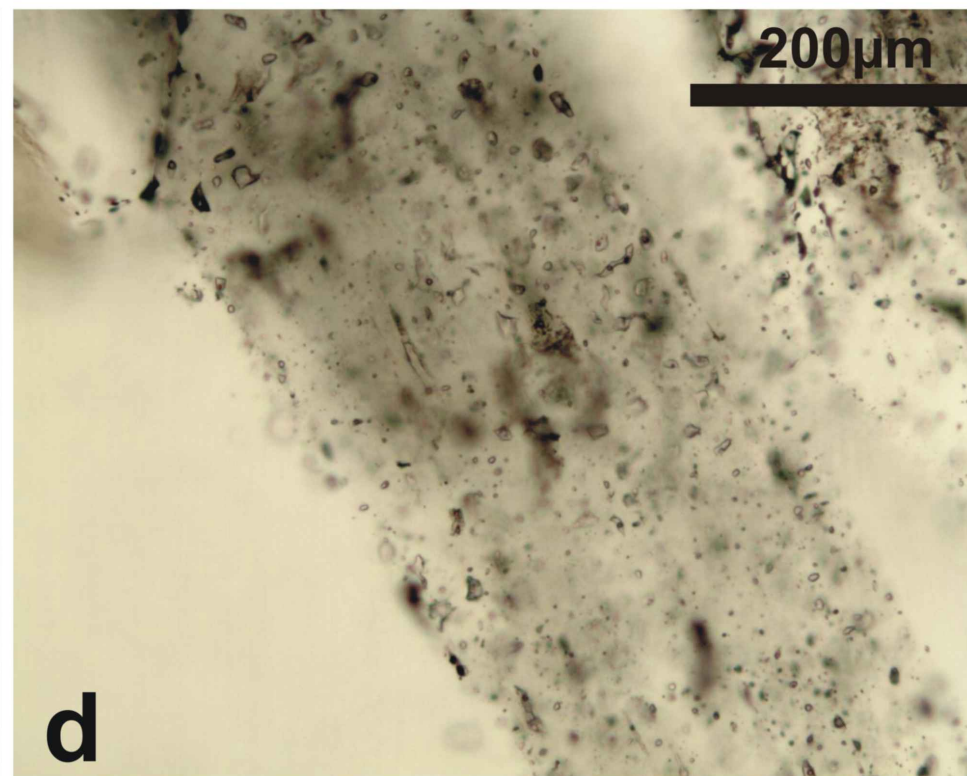
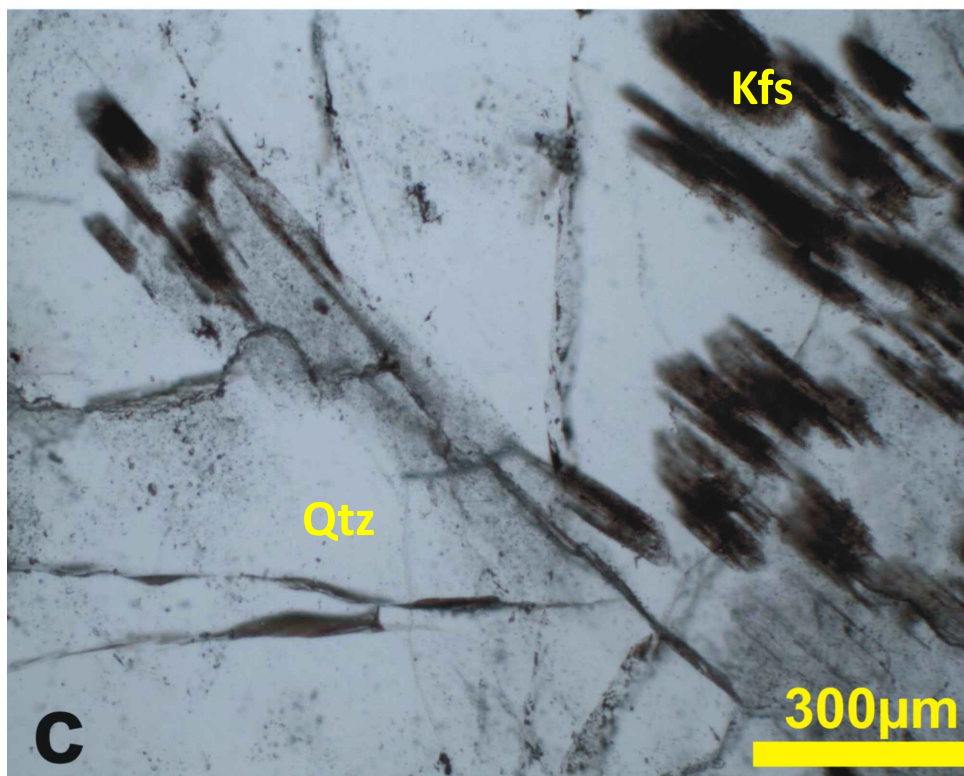


fluid inclusions restricted to pseudomorphs



pseudomorphs defined by fluid inclusions

Type 2



T Zone

Growth zones in quartz

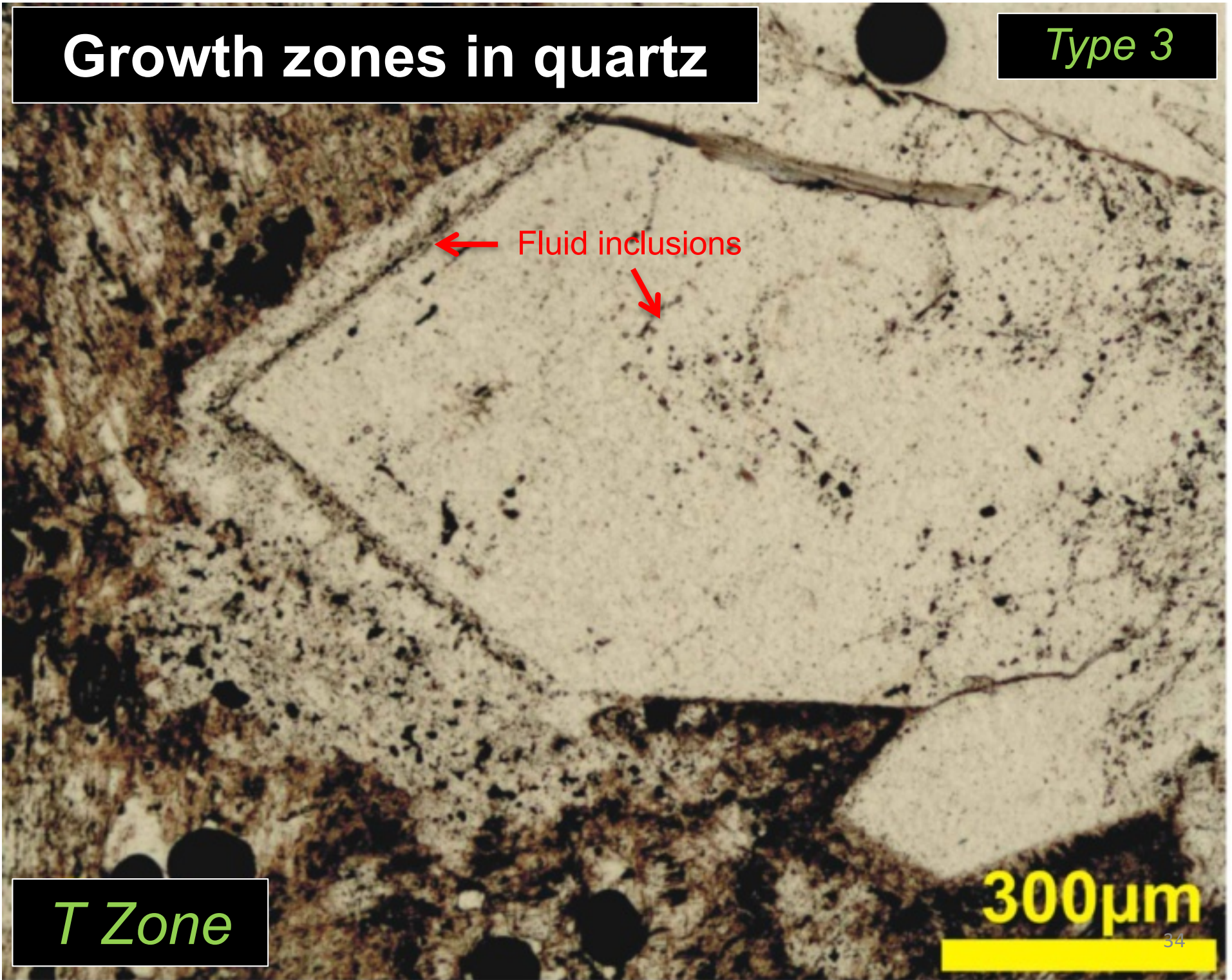
Type 3

← Fluid inclusions



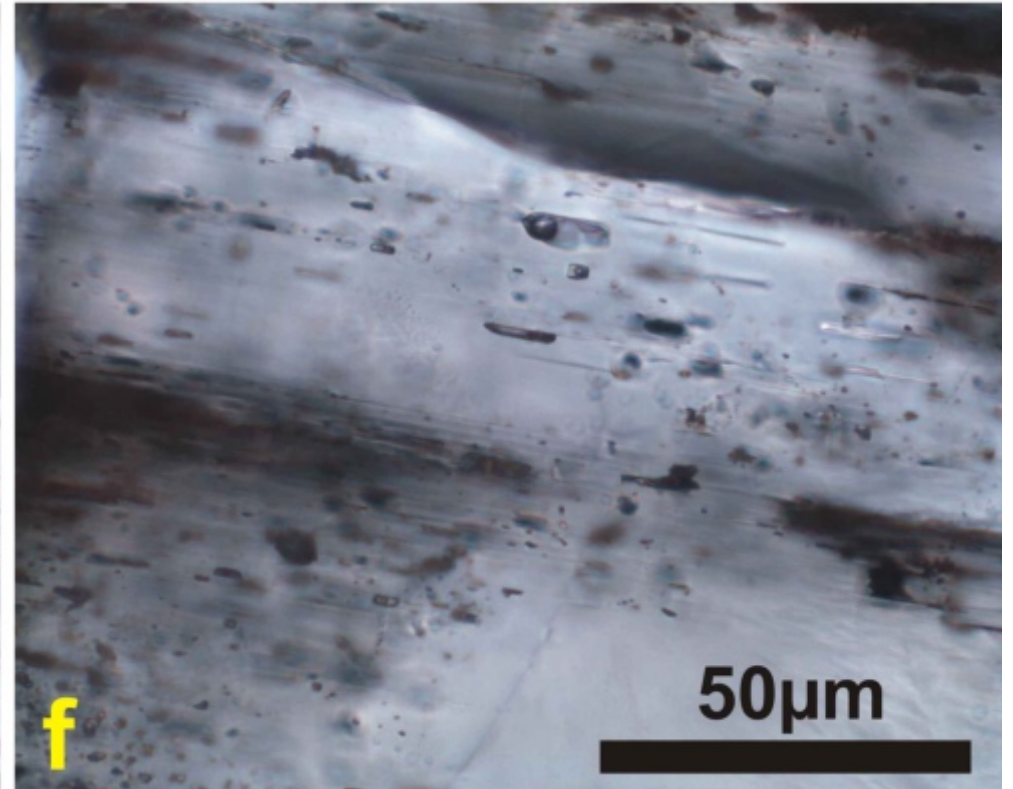
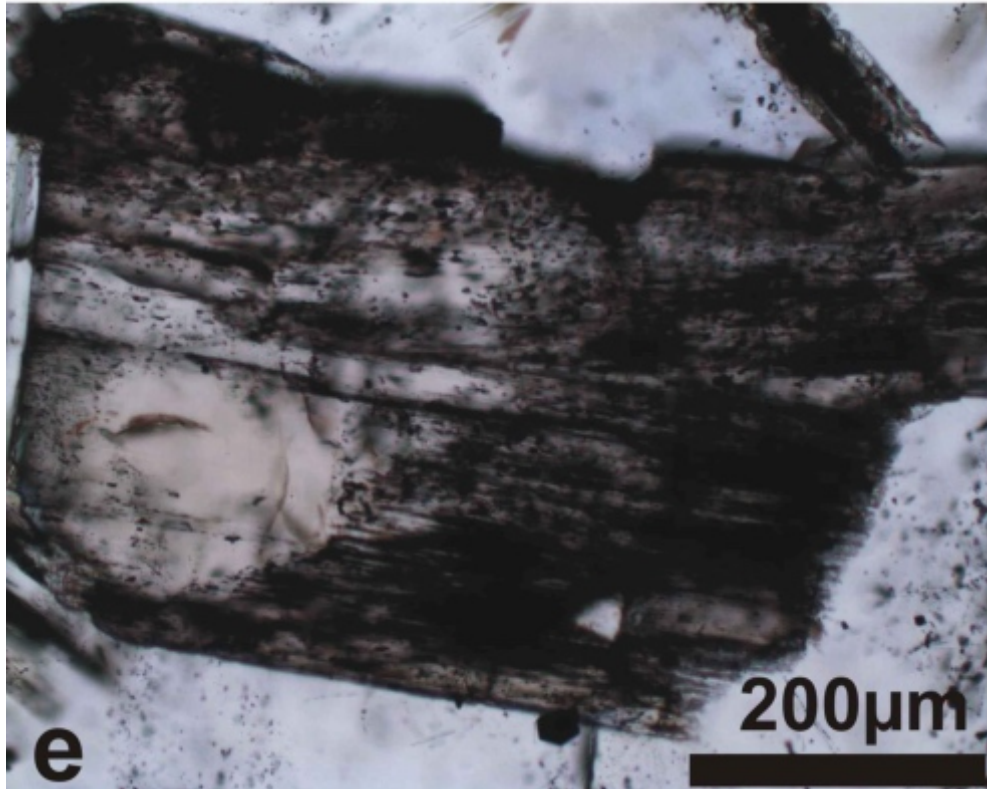
T Zone

300µm

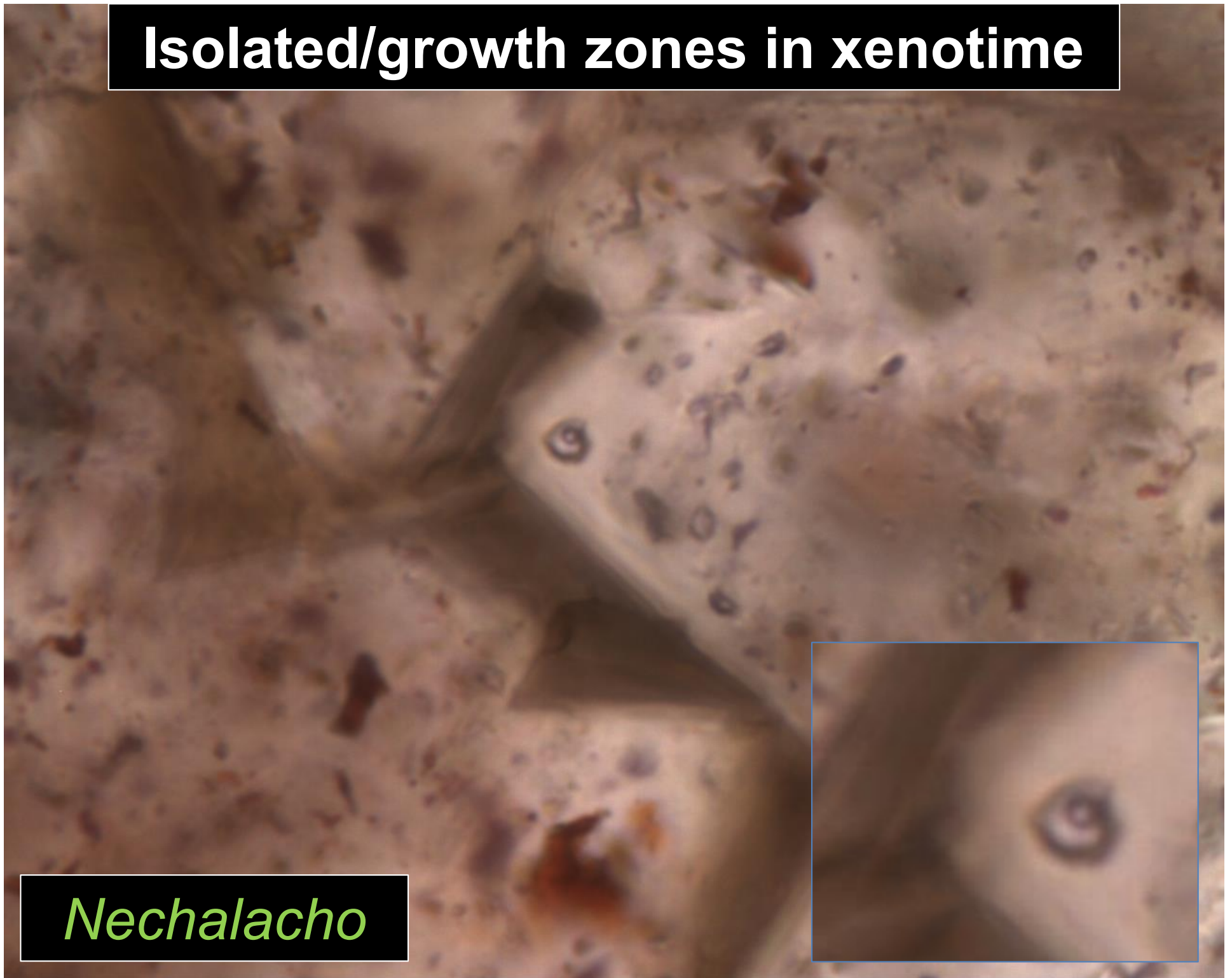


Oriented inclusions in bastnäsite

Type 6



Isolated/growth zones in xenotime



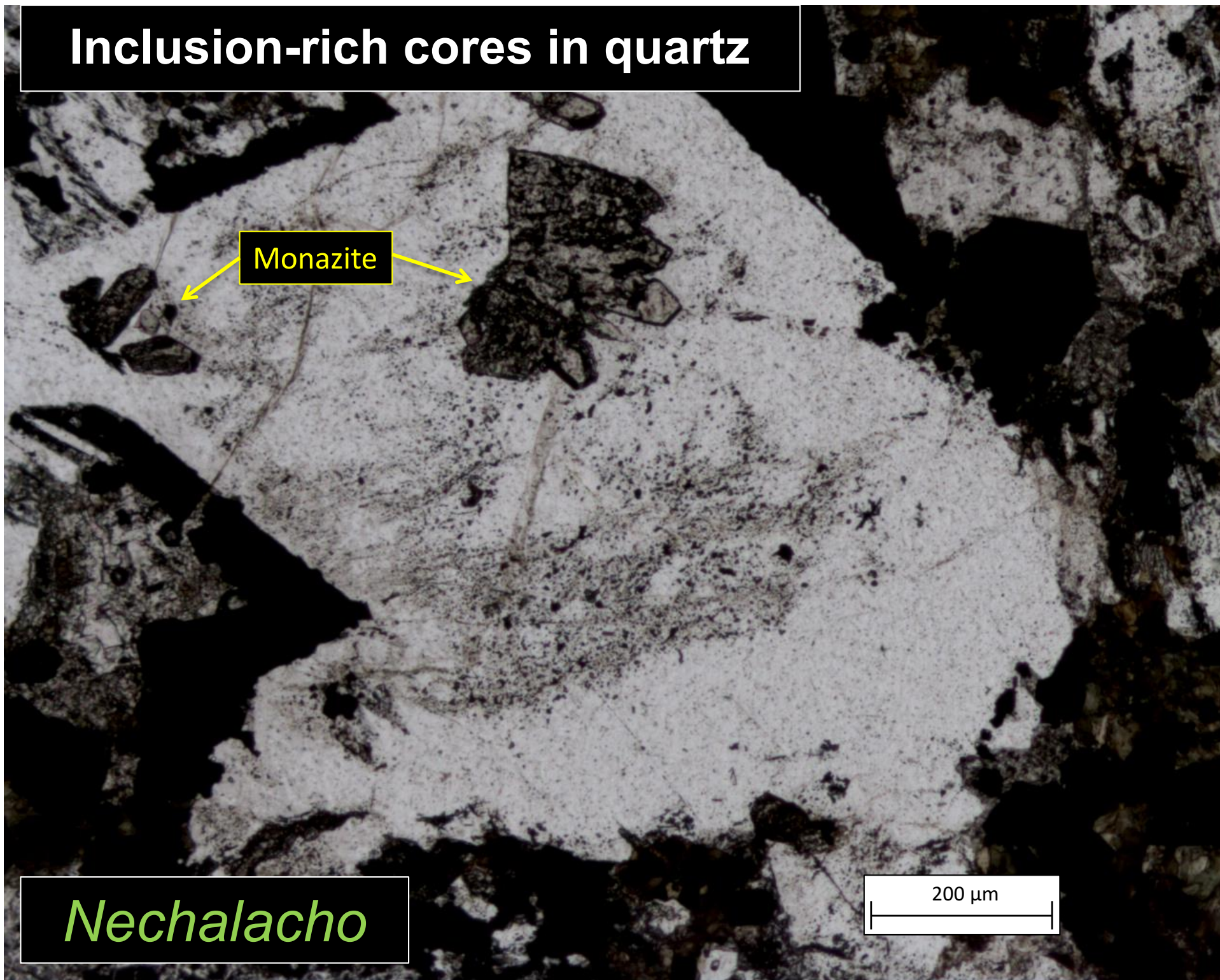
Nechalacho

Inclusion-rich cores in quartz

Monazite

Nechalacho

200 μm



Fluid Inclusion Assemblage (FIA) Classification

Pseudomorphs in
quartz

Growth Zones in *quartz*
(rarely in *fluorite*)

Isolated Inclusions in
xenotime + quartz

Elongate inclusions
parallel to the c-axis of
bastnäsite



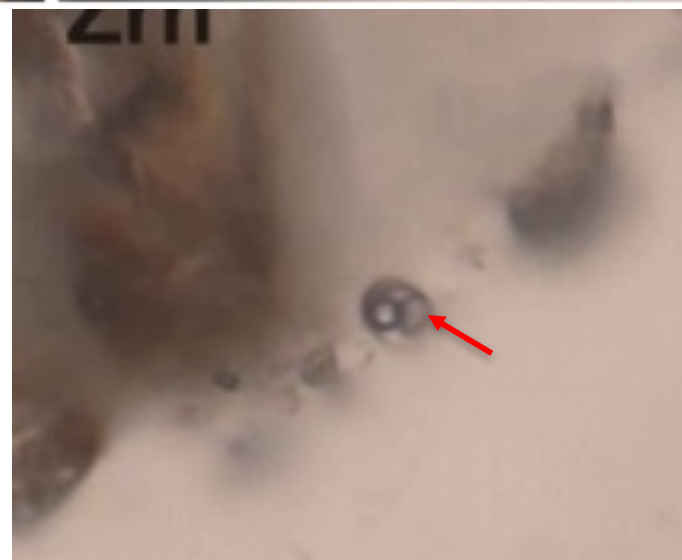
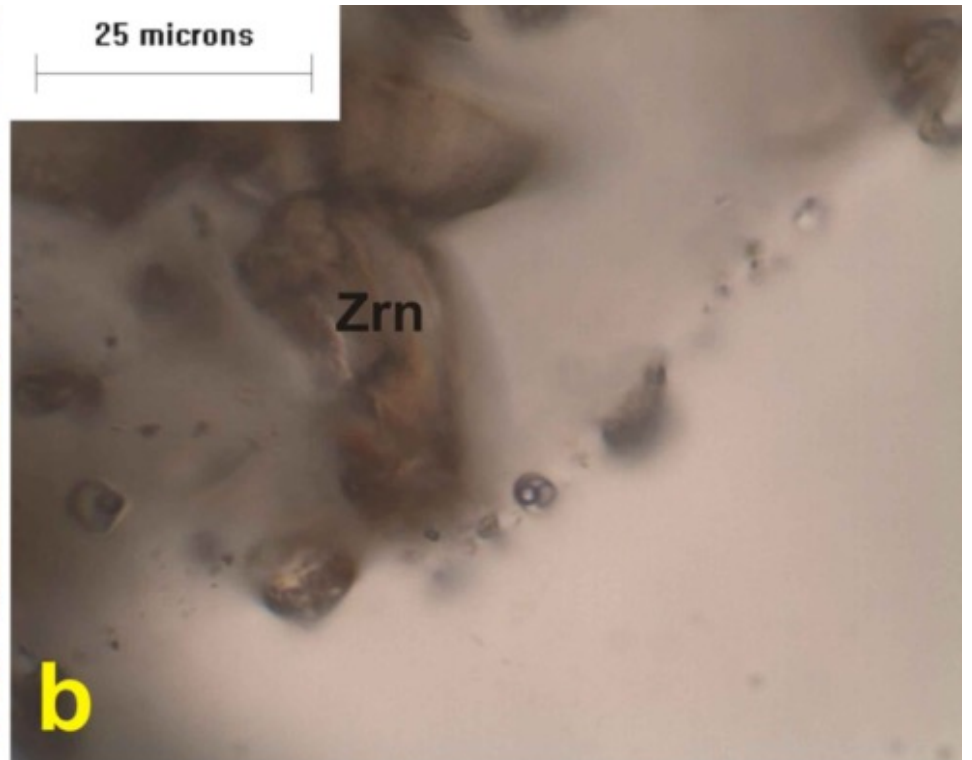
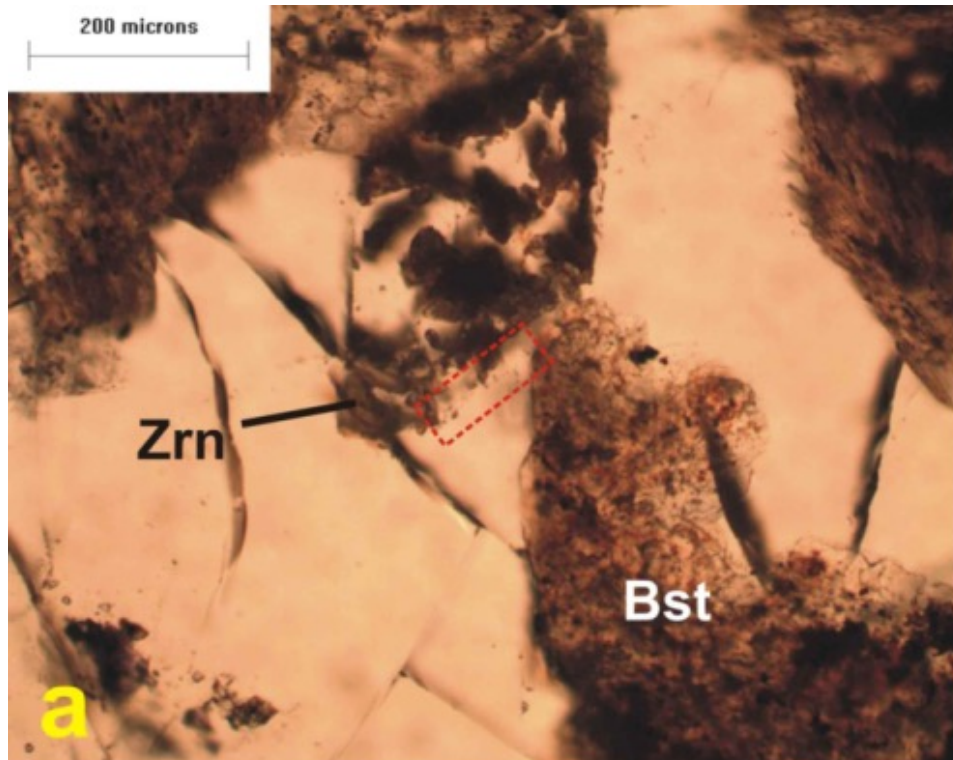
Aqueous, Liquid-rich
Liquid-Vapor \pm Solid (LV \pm S)

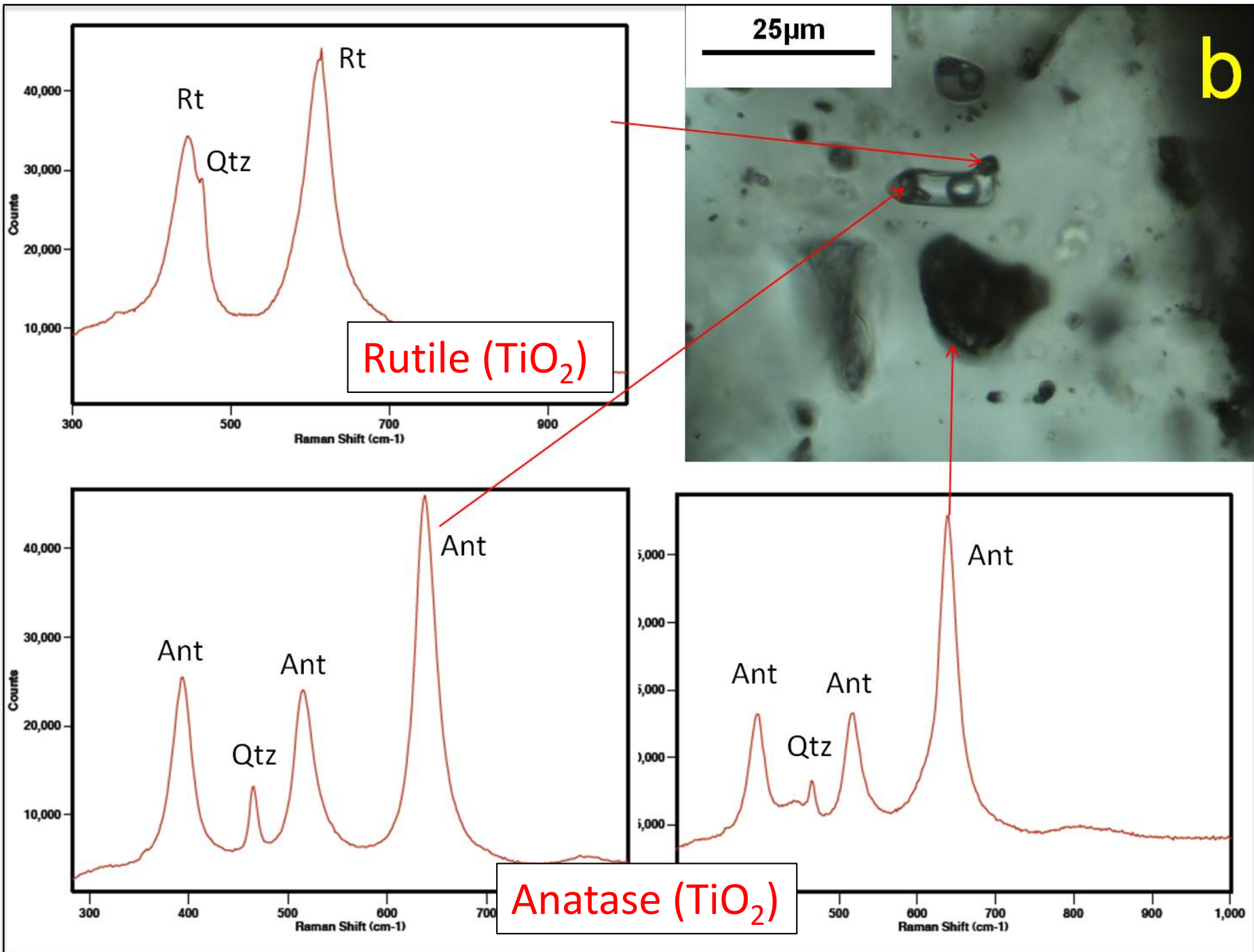


Very minor CO₂ and CH₄

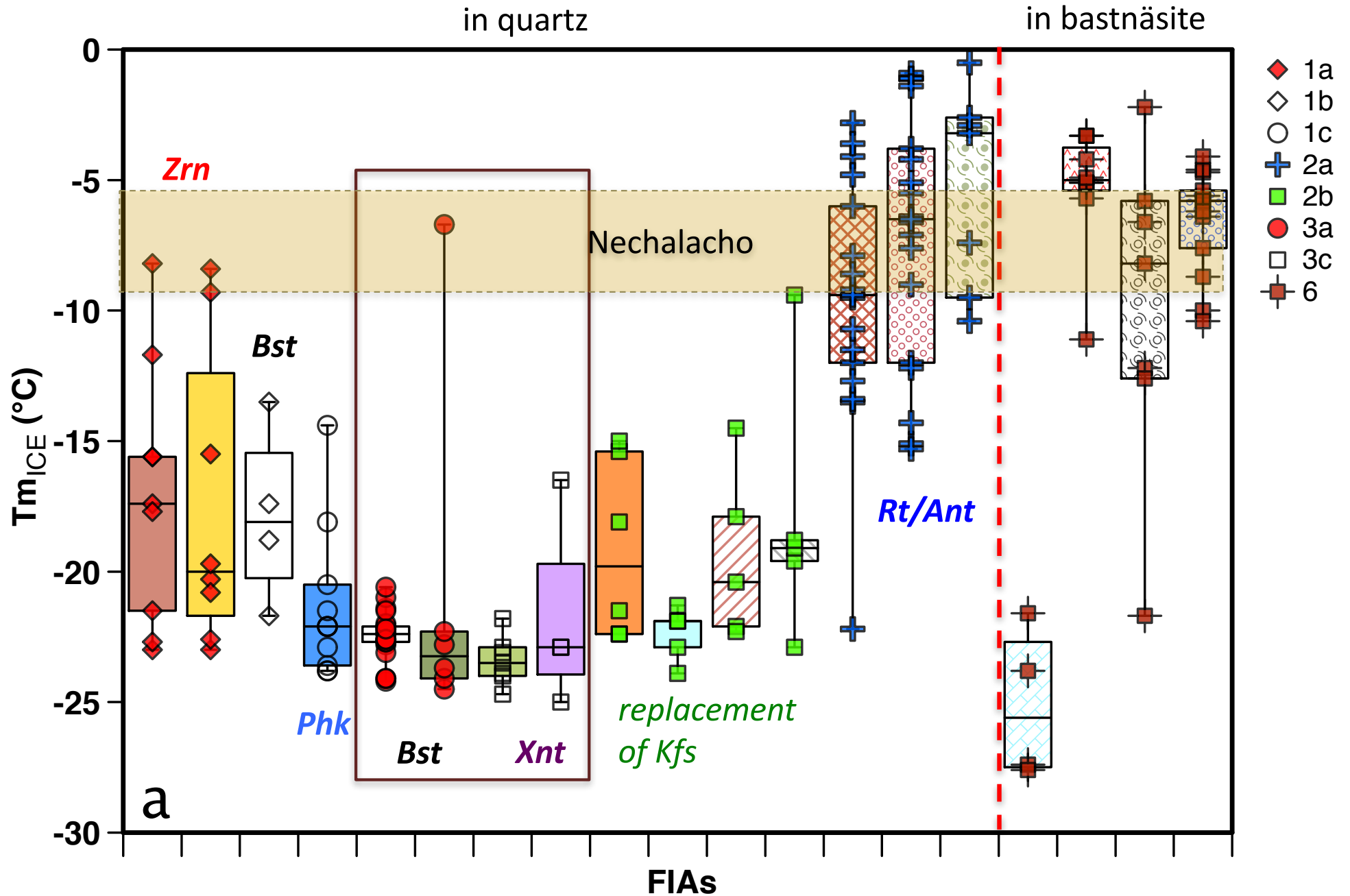
Trapped = zircon, anatase, rutile + ?

Trapped zircon in fluid inclusions

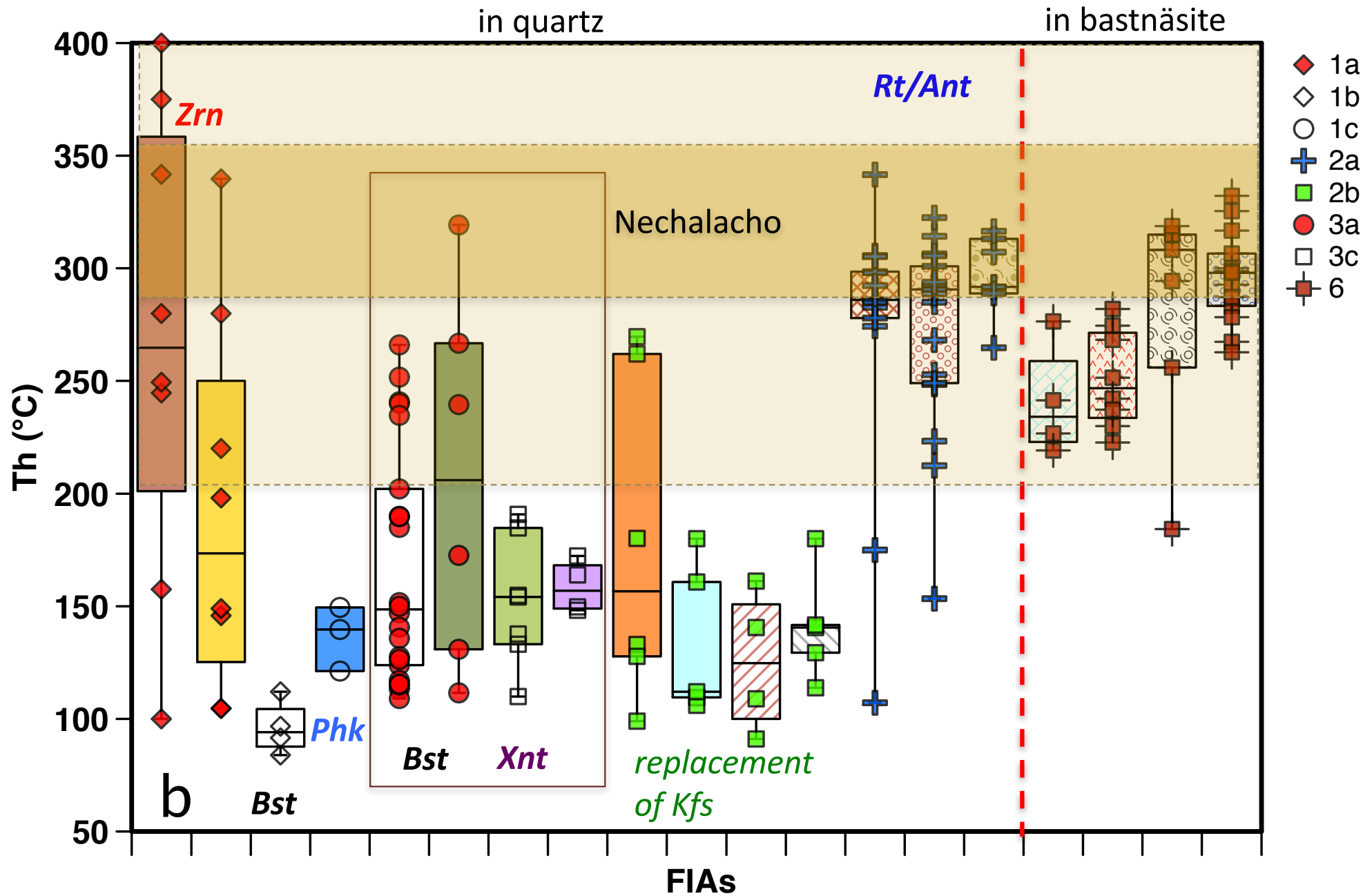




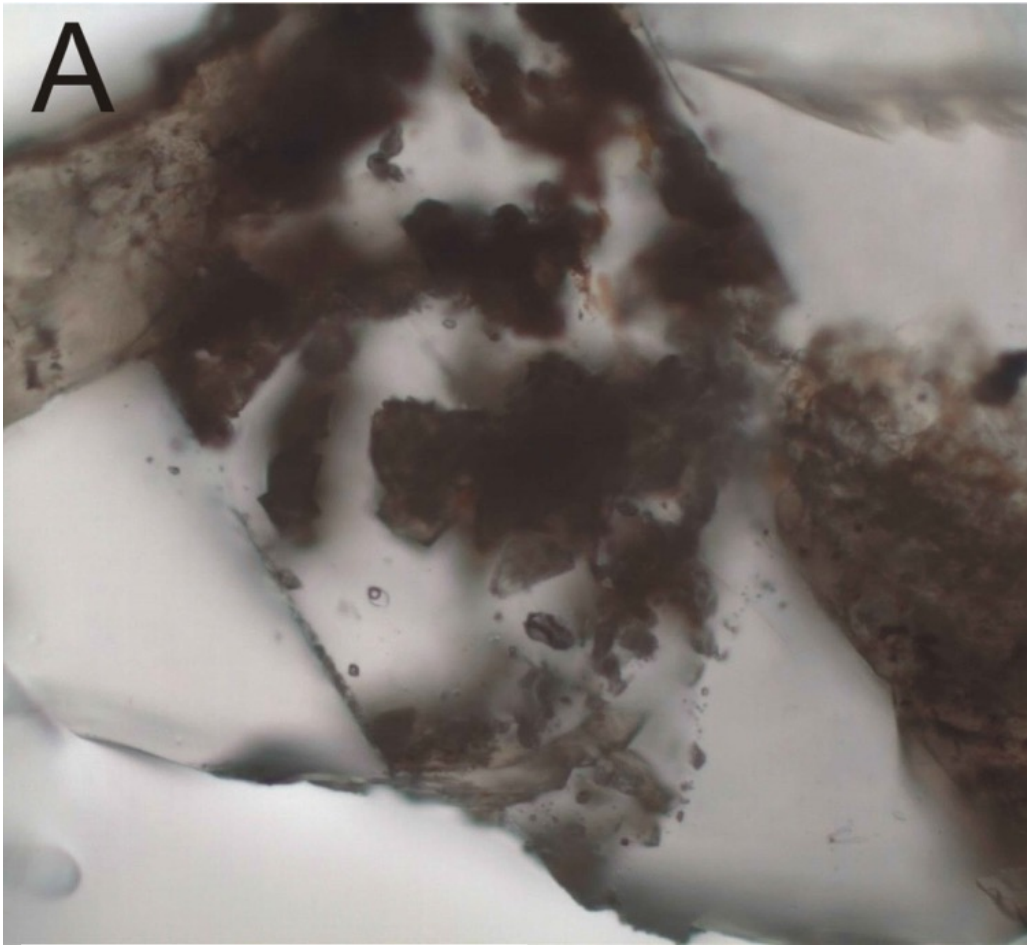
Ice melting temperatures



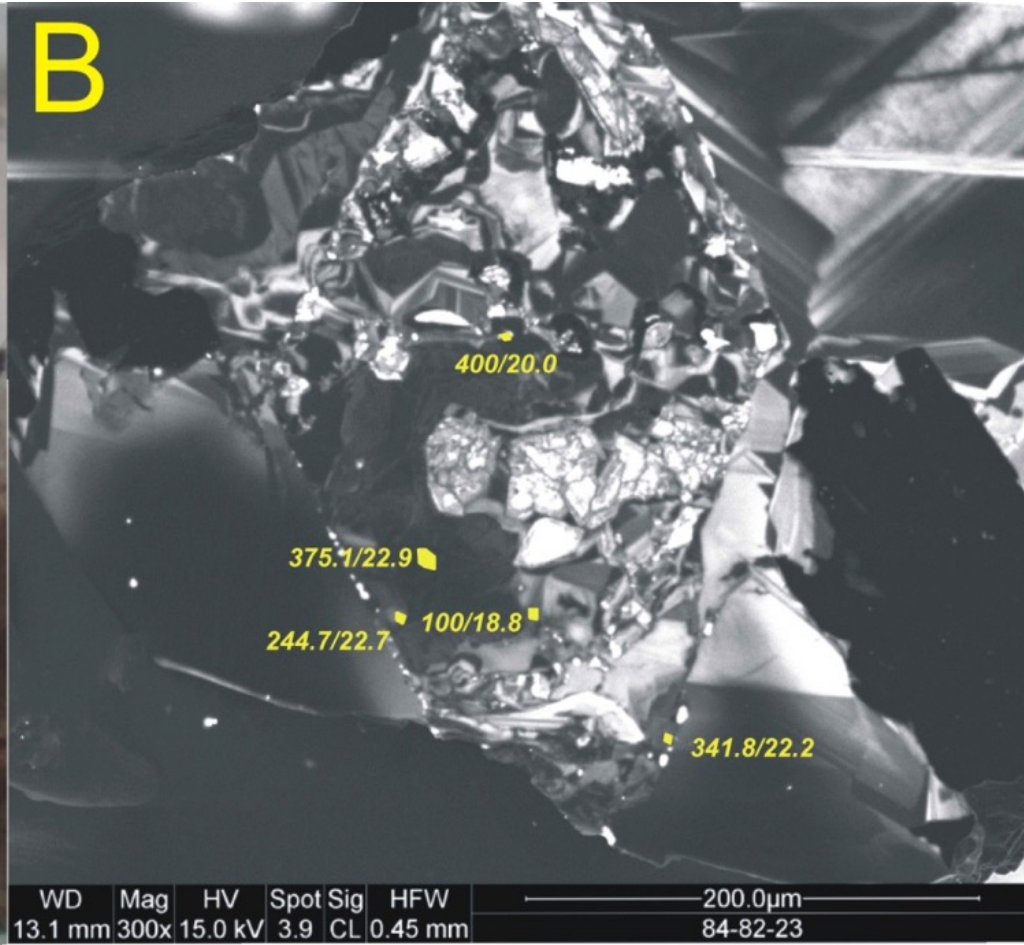
Homogenization temperatures: T_h LV \rightarrow L



Complex growth and fluid history

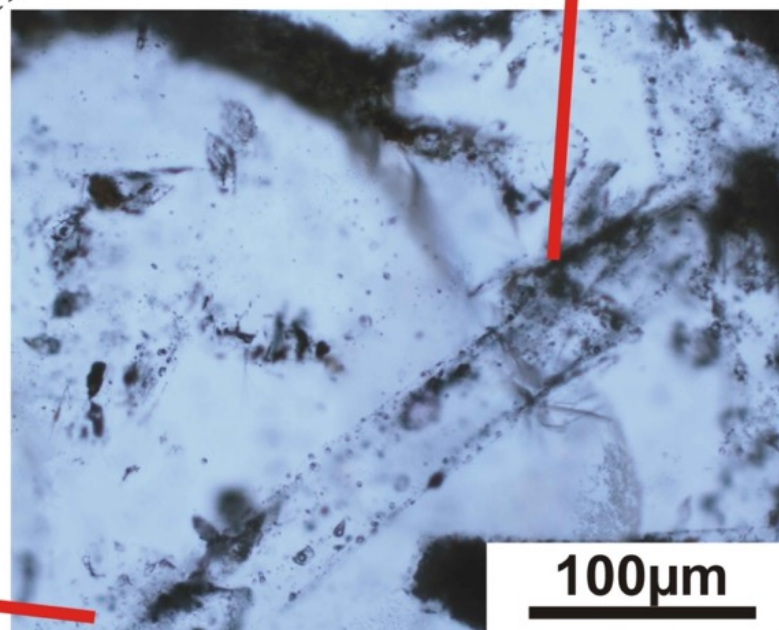
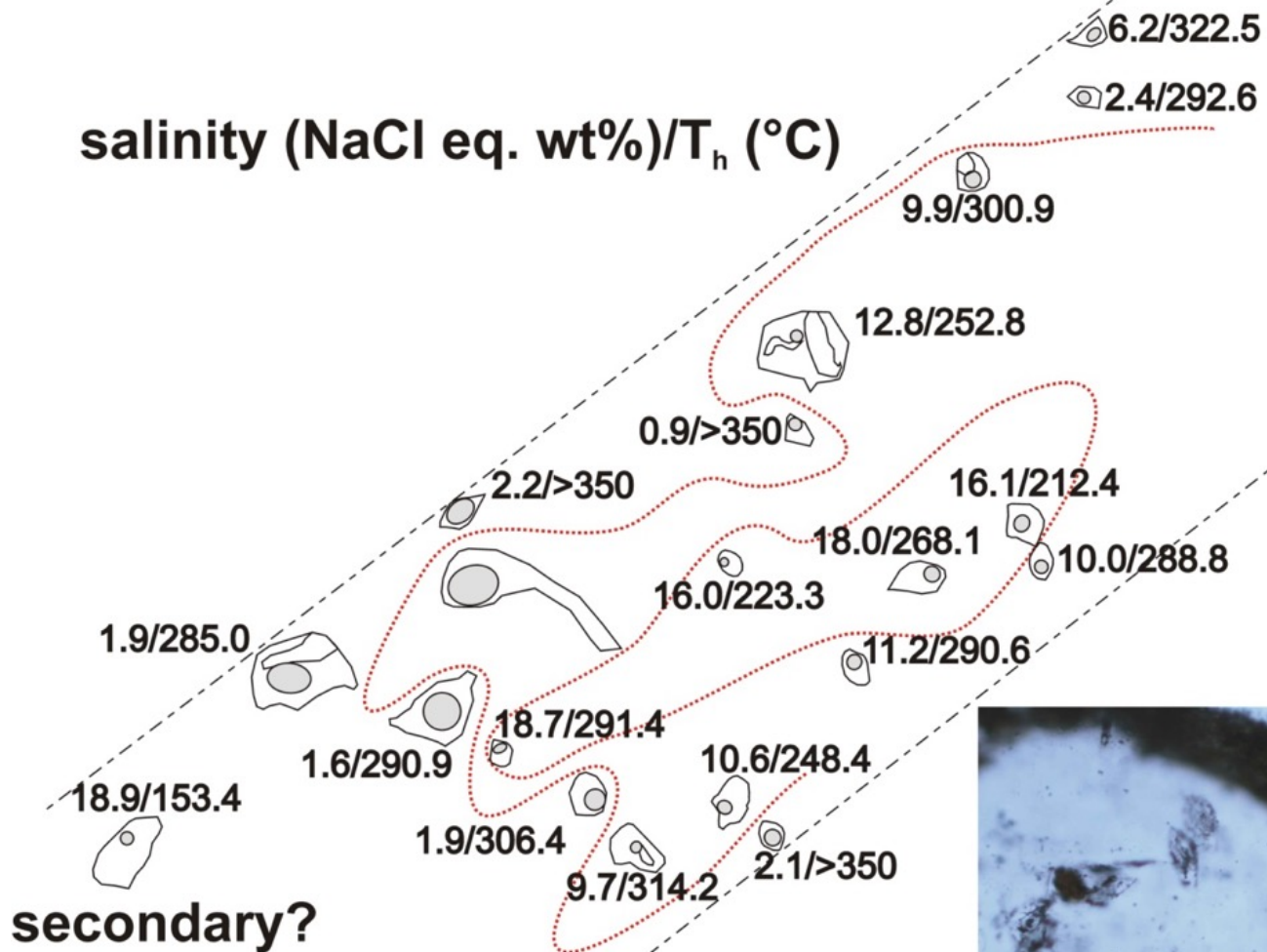


transmitted light

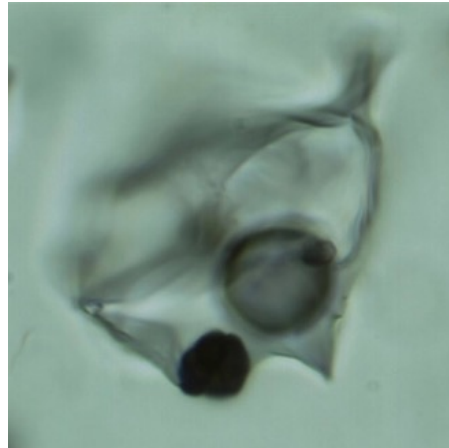


cathodoluminescence

Evolution of T and salinity



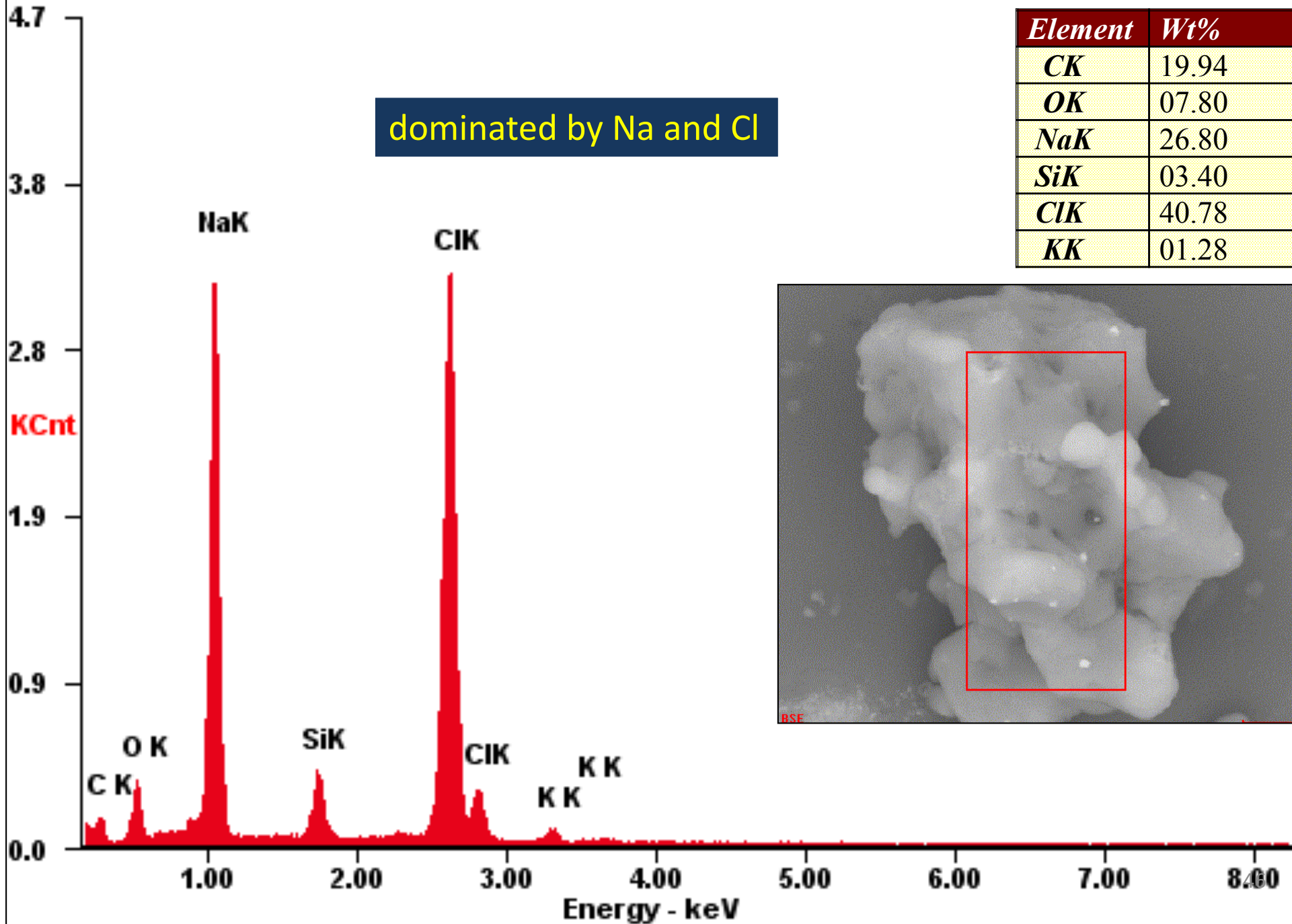
What about fluid chemistry?



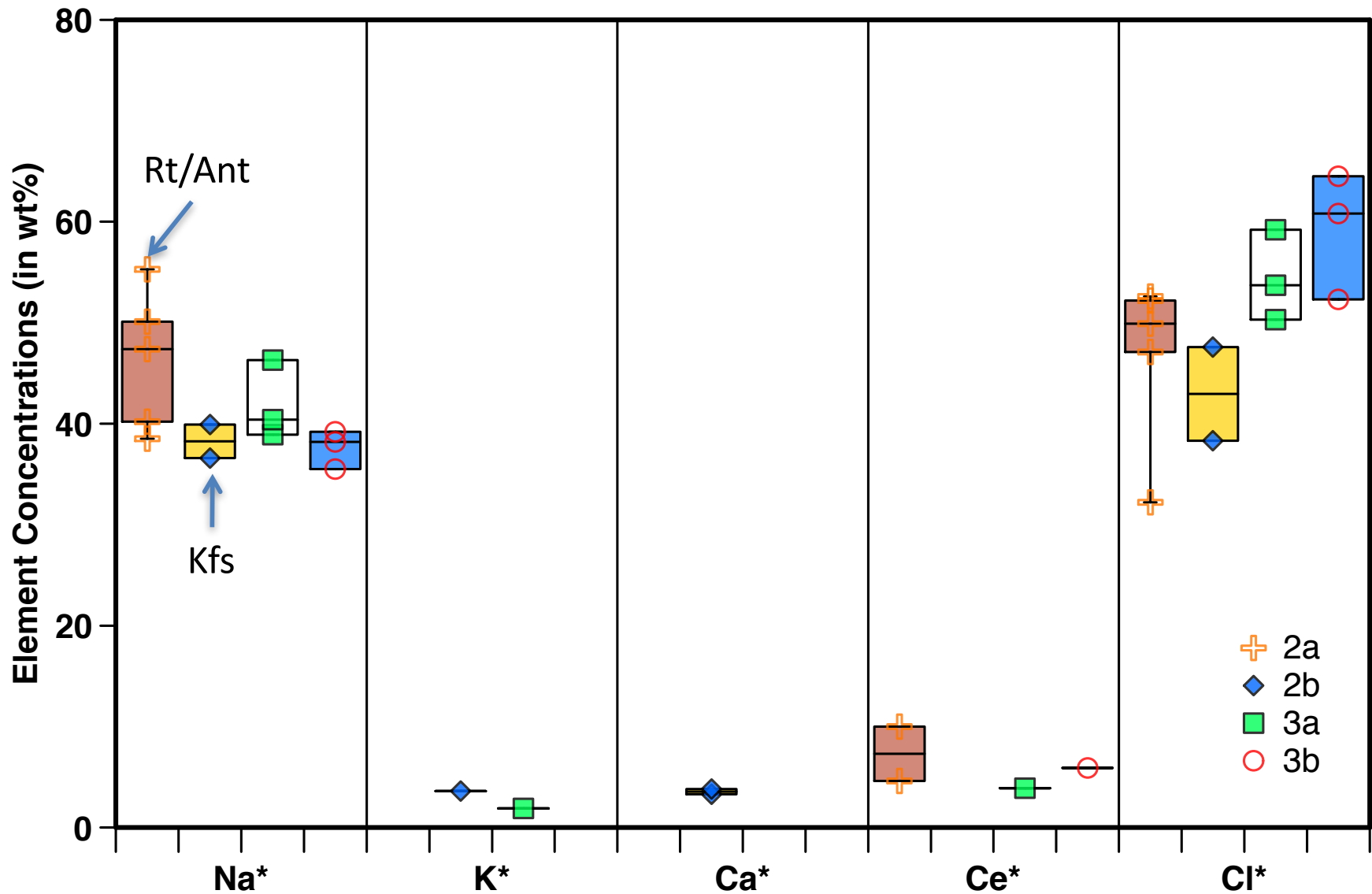
Energy-dispersive spectroscopy of decrepitates

dominated by Na and Cl

<i>Element</i>	<i>Wt%</i>
<i>CK</i>	19.94
<i>OK</i>	07.80
<i>NaK</i>	26.80
<i>SiK</i>	03.40
<i>ClK</i>	40.78
<i>KK</i>	01.28

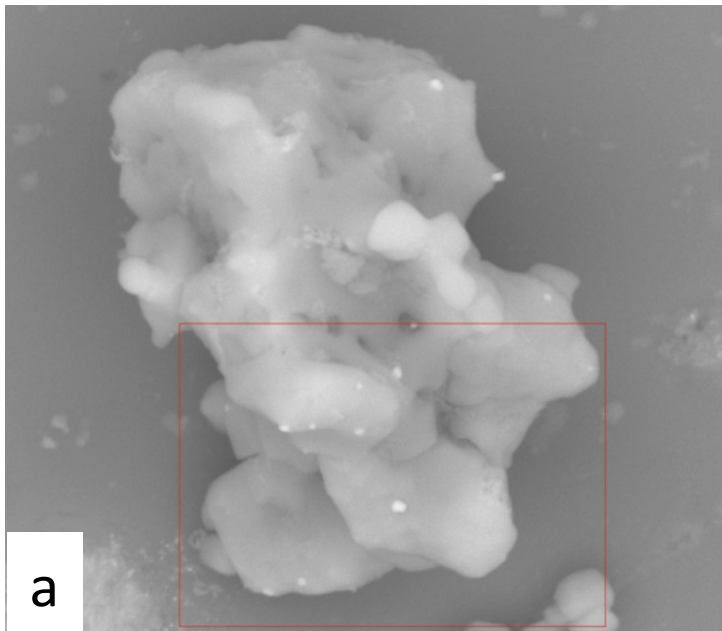


EDS analysis of decrepitates

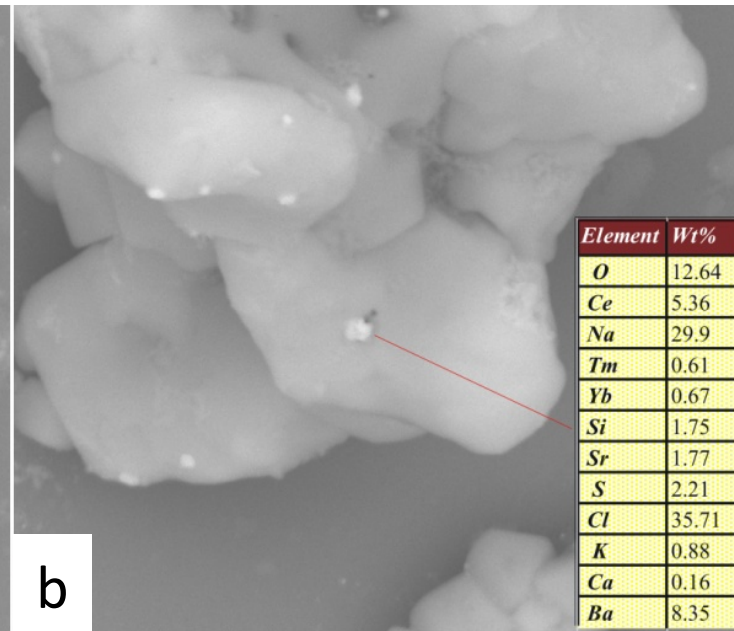


other elements: S, Ba, Sr, Ti, Th, other REE

Little F

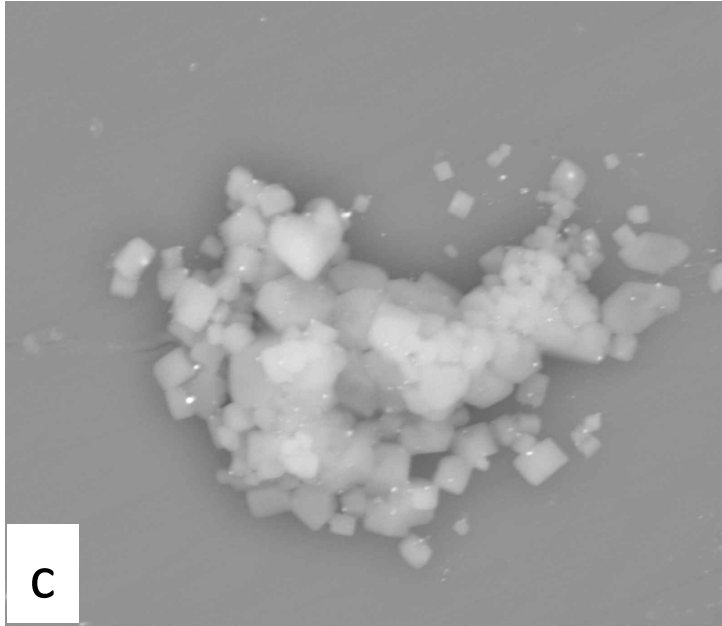


WD 10.7 mm Mag 7512x HV 15.0 kV Spot 3.9 HFW 18.00 µm Sig BSE 5.0µm NTZ7B

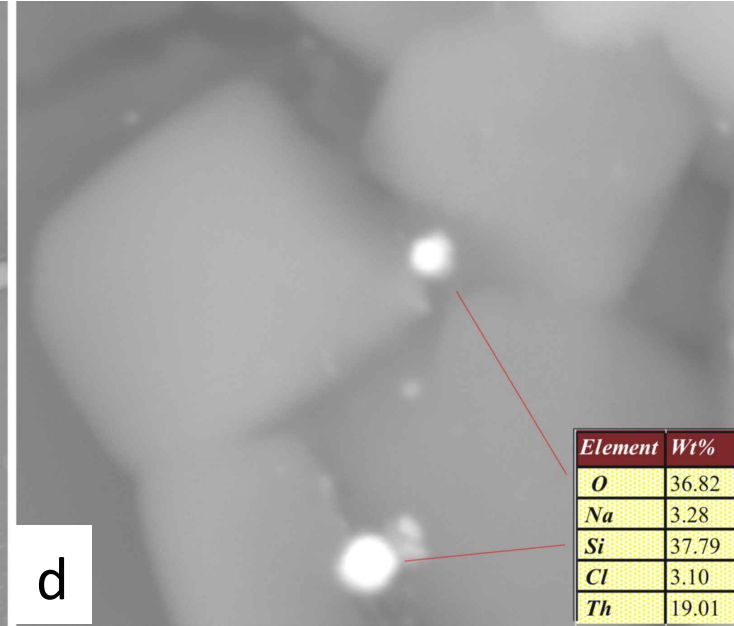


WD 10.7 mm Mag 13000x HV 15.0 kV Spot 3.9 HFW 10.40 µm Sig BSE 2.0µm NTZ7B

Element	Wt%
<i>O</i>	12.64
<i>Ce</i>	5.36
<i>Na</i>	29.9
<i>Tm</i>	0.61
<i>Yb</i>	0.67
<i>Si</i>	1.75
<i>Sr</i>	1.77
<i>S</i>	2.21
<i>Cl</i>	35.71
<i>K</i>	0.88
<i>Ca</i>	0.16
<i>Ba</i>	8.35



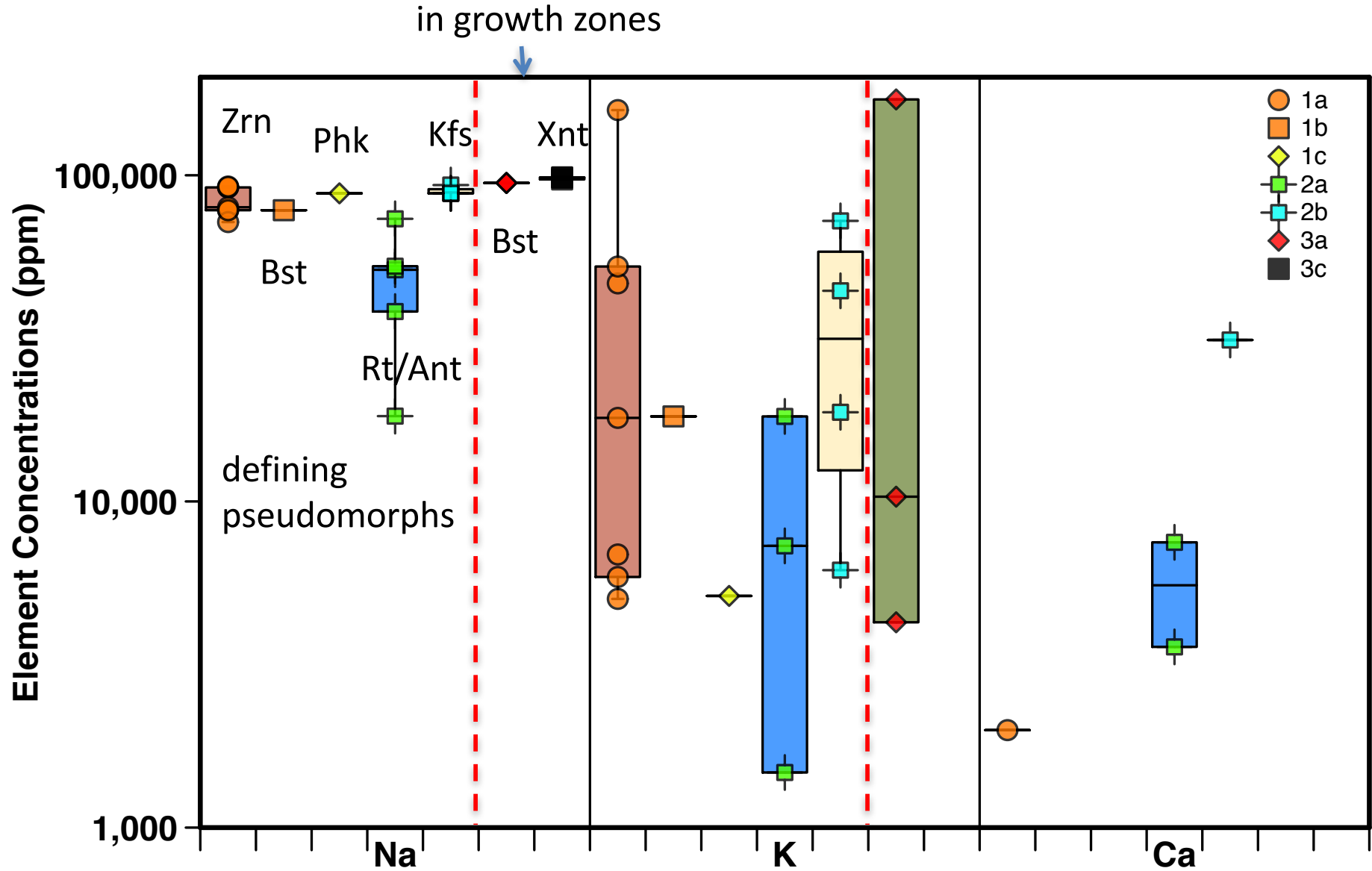
WD 10.7 mm Mag 8000x HV 15.0 kV Spot 3.9 HFW 16.90 µm Sig BSE 5.0µm IS11-04



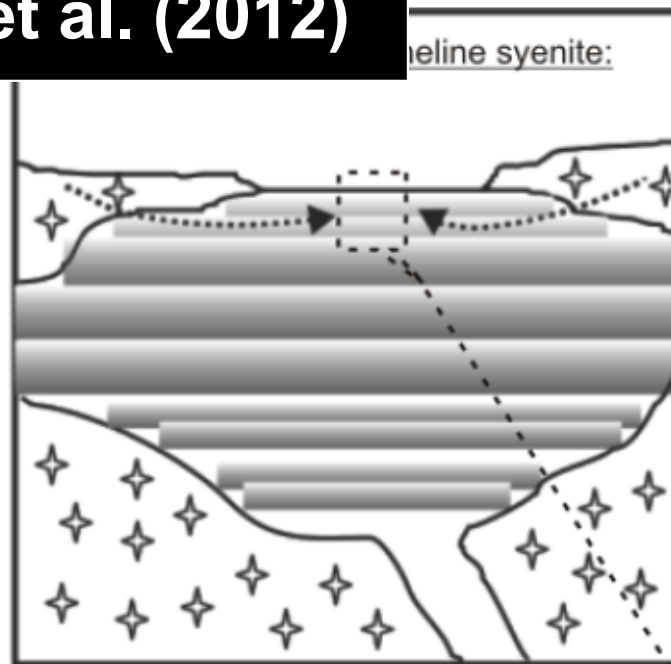
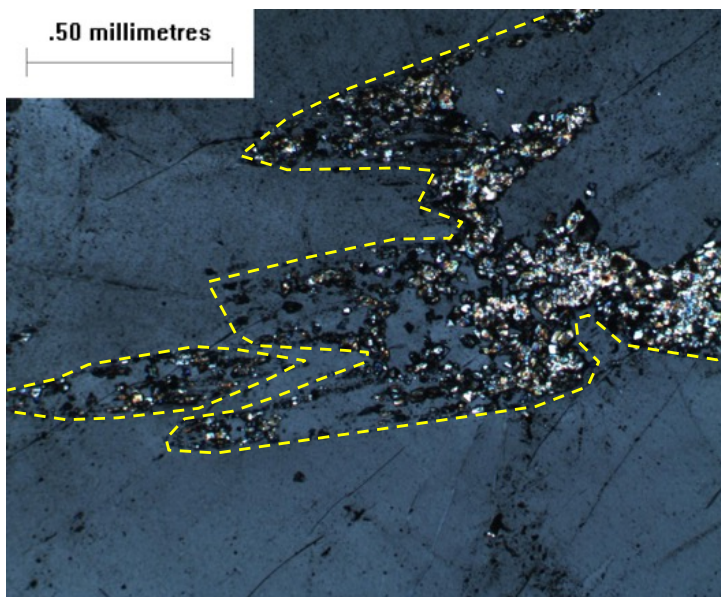
WD 10.7 mm Mag 73667x HV 15.0 kV Spot 3.9 HFW 1.84 µm Sig BSE 500.0nm IS11-04

Element	Wt%
<i>O</i>	36.82
<i>Na</i>	3.28
<i>Si</i>	37.79
<i>Cl</i>	3.10
<i>Th</i>	19.01

LA-ICP-MS analysis of fluid inclusions: Na, K, Ca

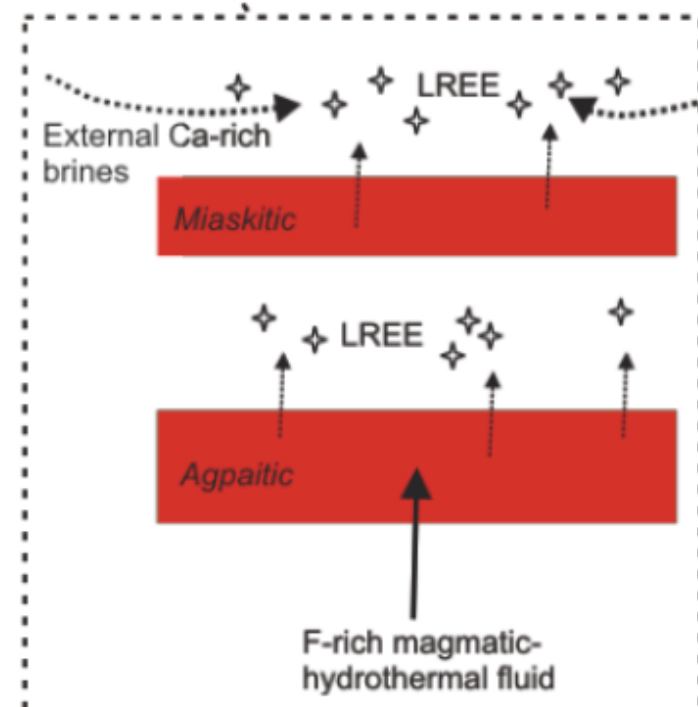


Mixing Model: Sheard et al. (2012)



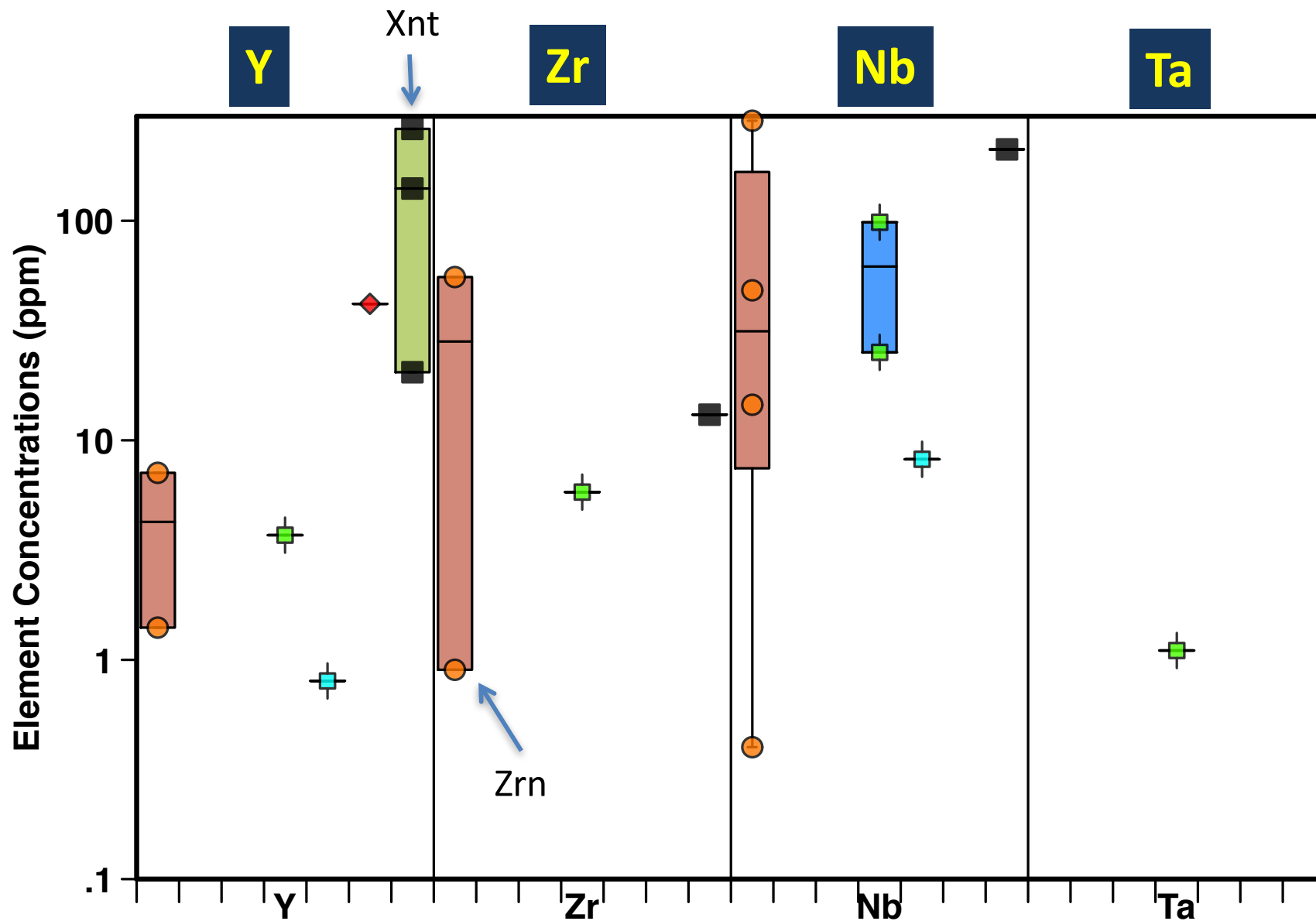
convective

Alteration & remobilization:

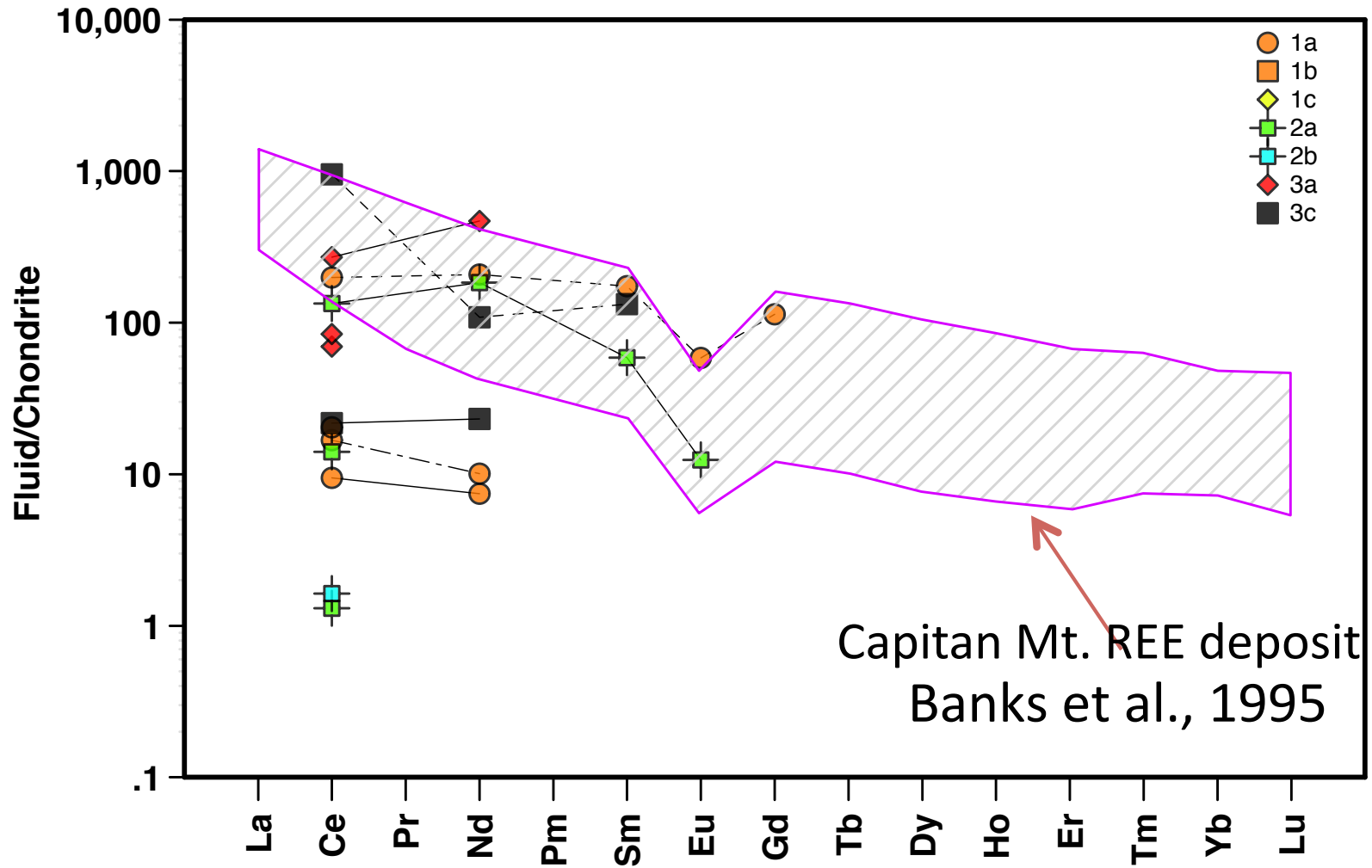


T Zone fluids: low F, Ca

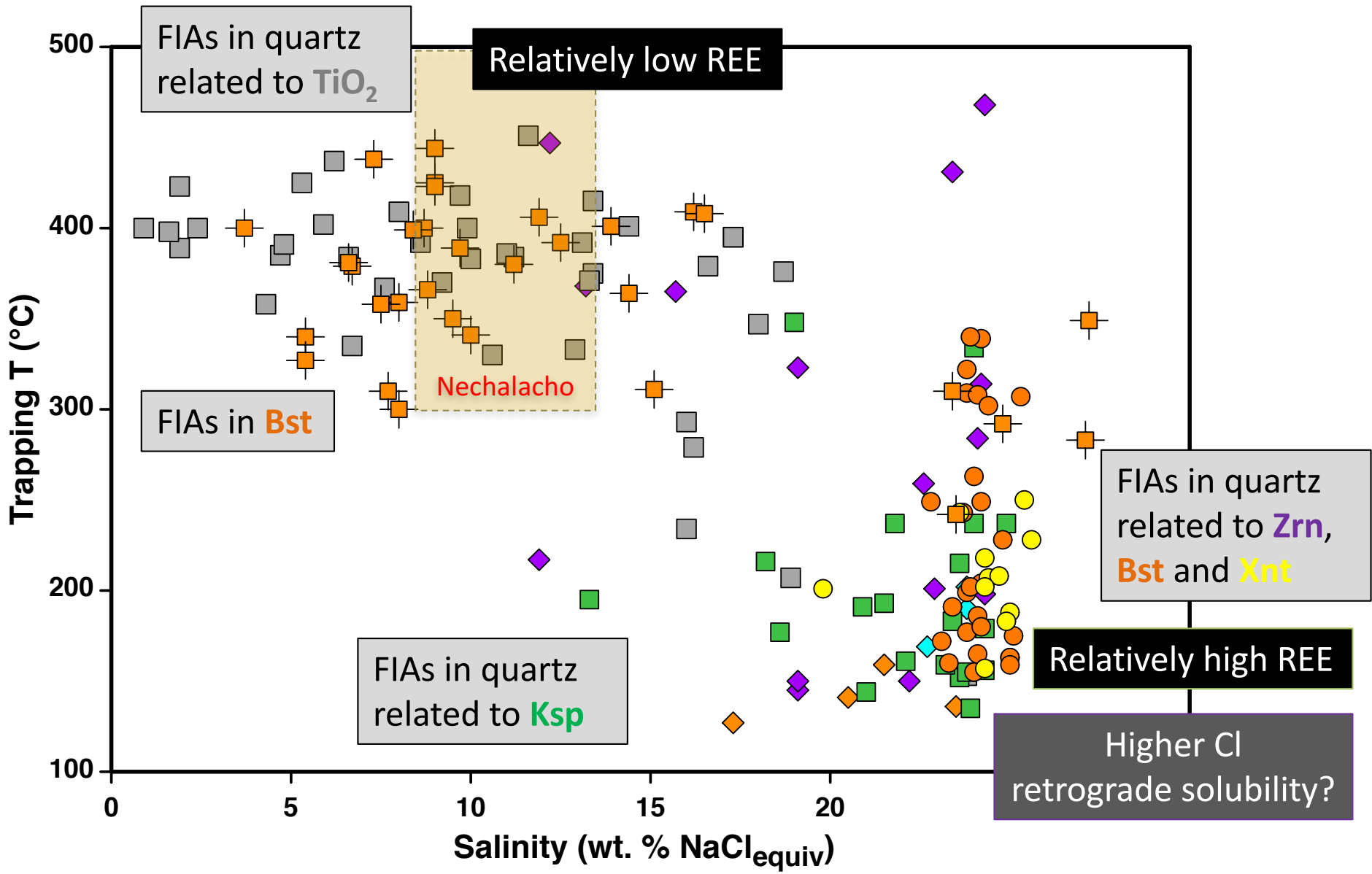
Rare-metal concentrations



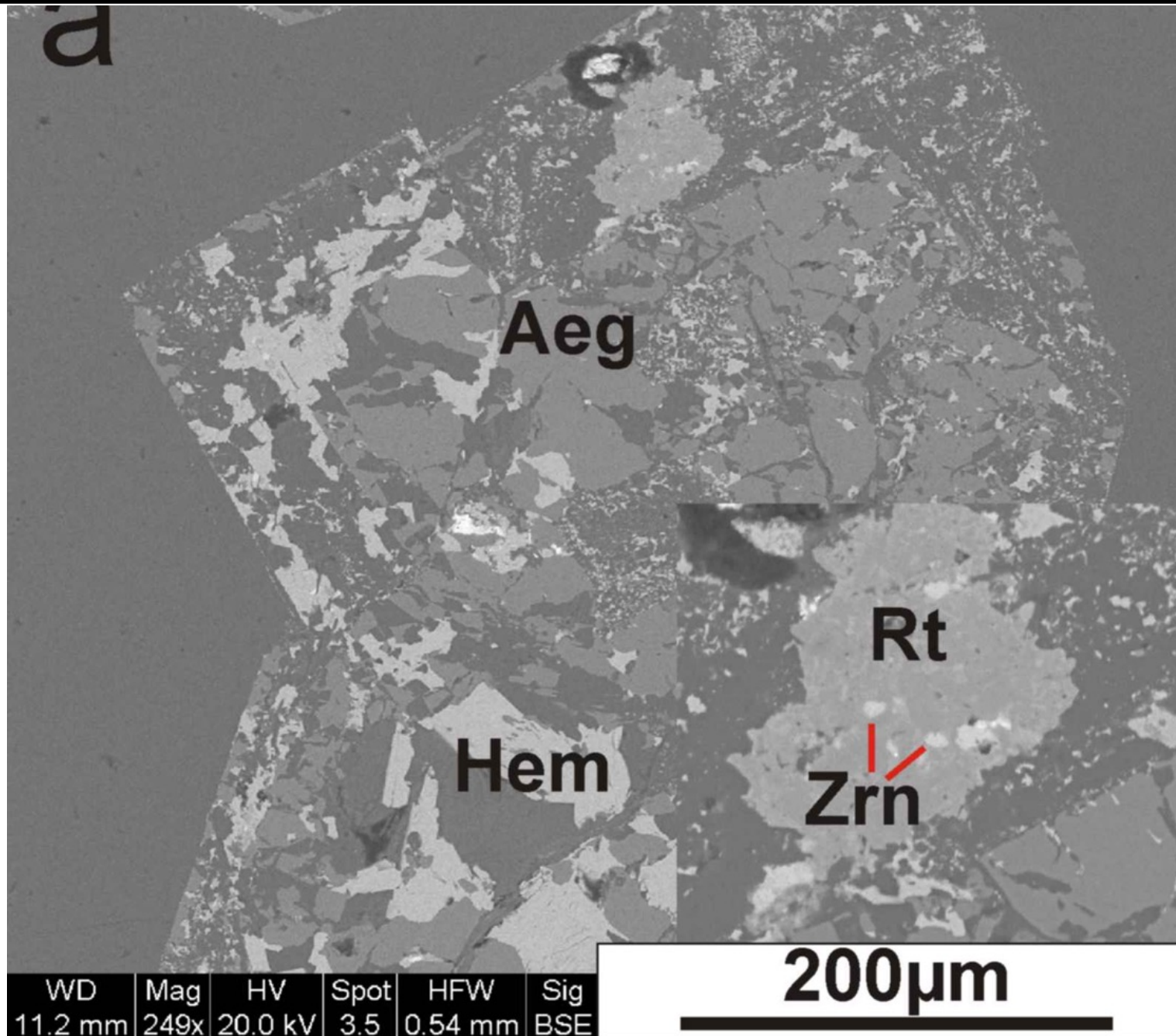
REE concentrations



Two Populations

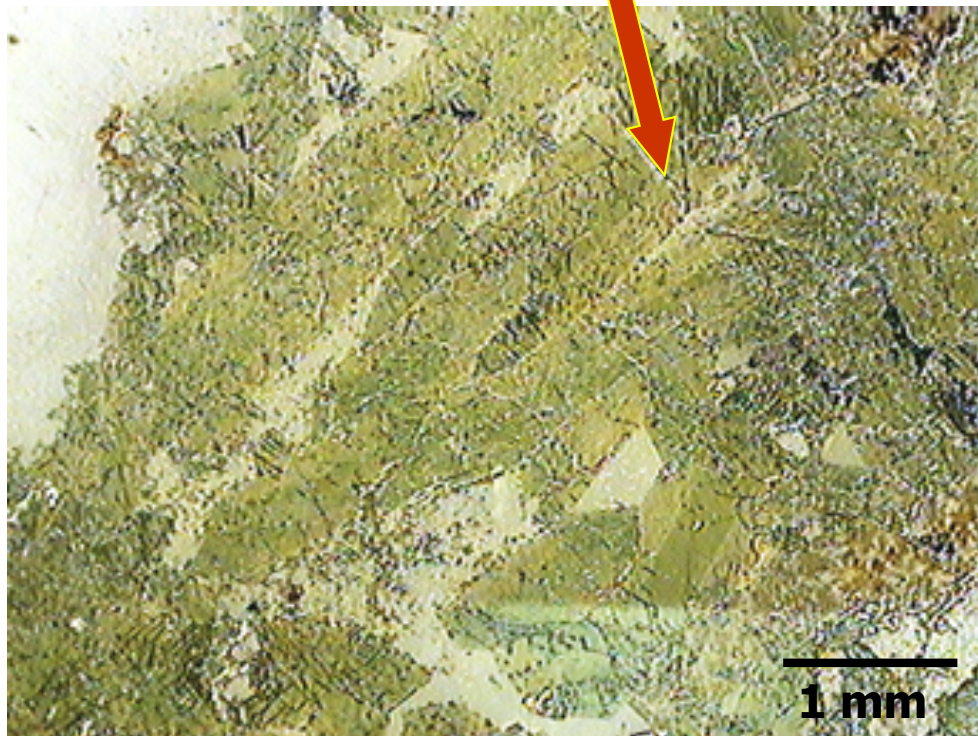


Reasons for Enrichment?: Aegirine and Mica replacement

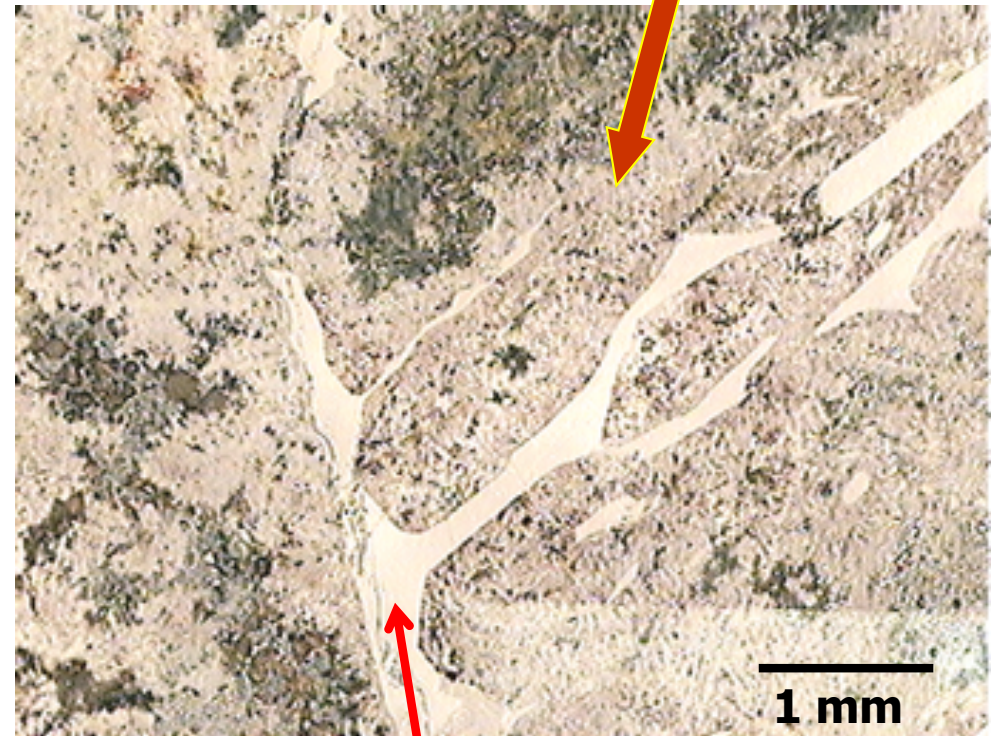


Strange Lake, Quebec/Labrador

aegirine



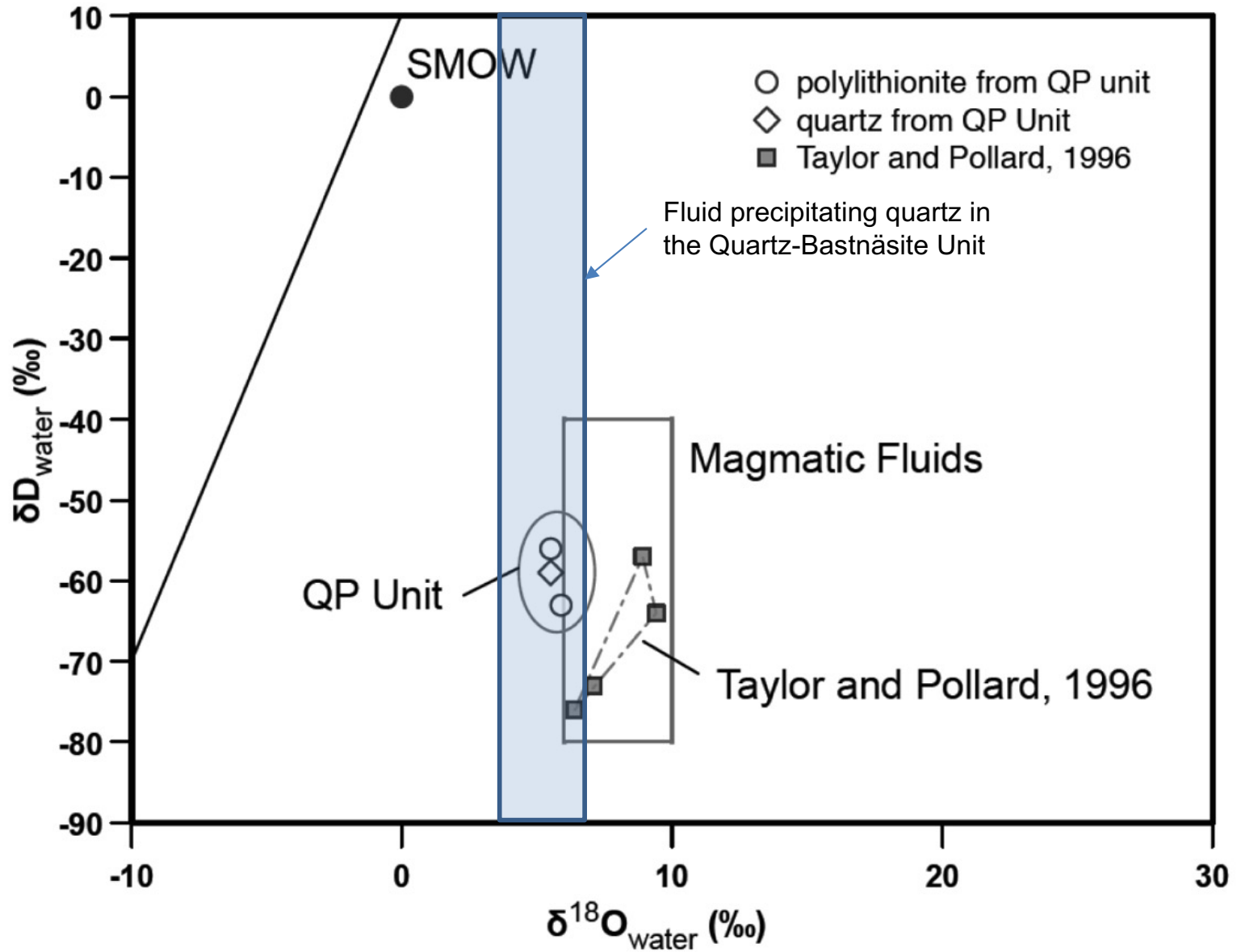
gittinsite + qtz + hm



qtz

Photos courtesy of J. Gagnon

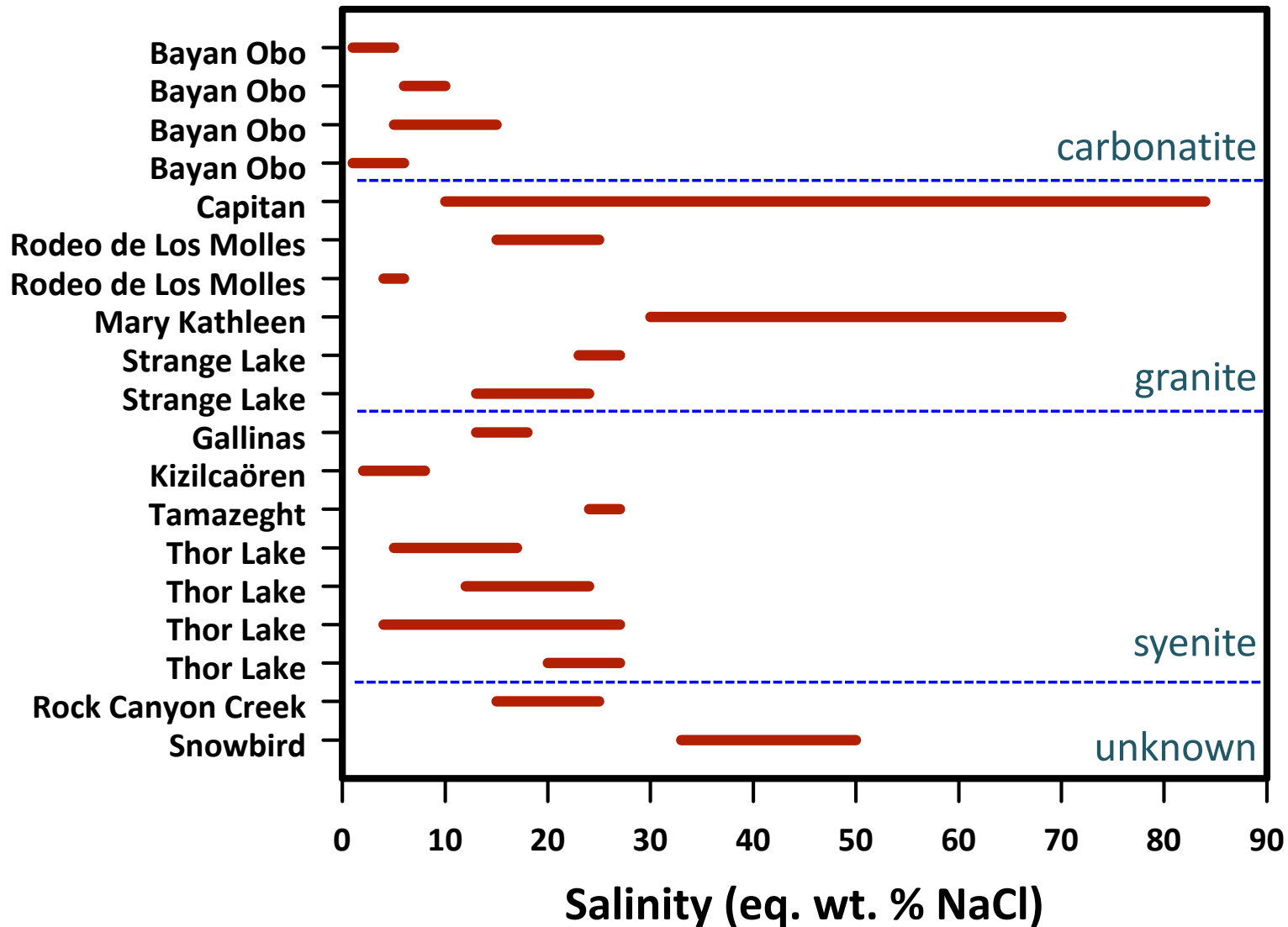
A magmatic source? Li and Be?



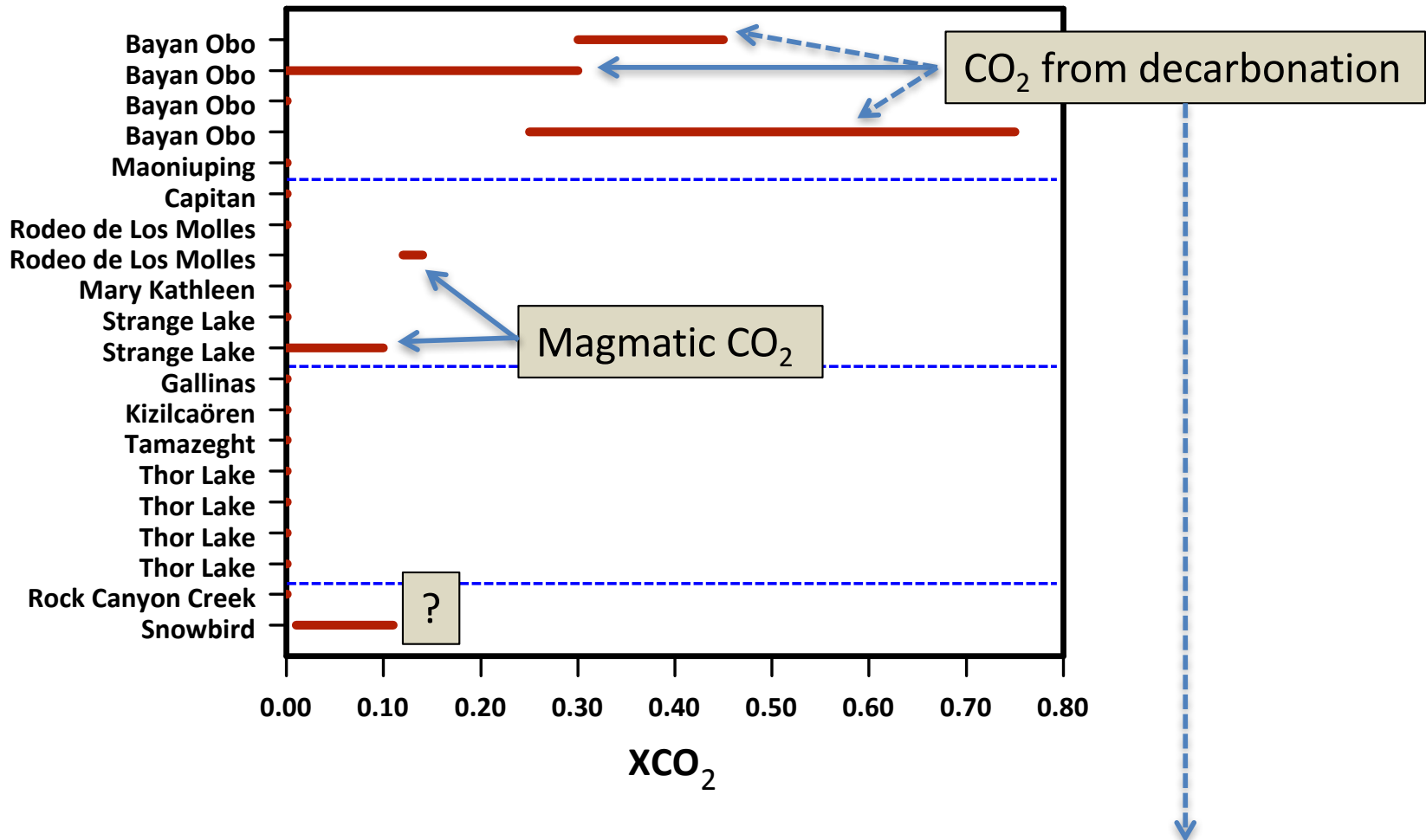
Questions: what roles do fluids play?

- Primary vs secondary concentration and enrichment?
 - Secondary at Thor Lake
- What types of fluids are capable of mobilizing significant rare metals?
 - Aqueous, low to moderate T and salinity
 - Low CO₂ and CH₄
- What is the evidence for this?
 - Ubiquitous replacement of primary minerals by hydrothermal assemblages
 - Primary fluid inclusions in pseudomorphs with rare-metal minerals, in rare-metal minerals, and in growth zones in quartz
- Are rare metals present in the fluids?
 - yes
- What concentrations?
 - 10s → 1000s ppm
 - Highest concentrations: ~ 150 to 250 °C and from ~ 20 to 25 wt. % NaCl_{equiv}
- What controls enrichment?
 - Replacement of mafic minerals and micas

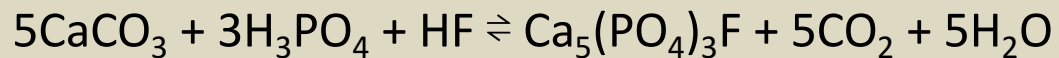
Salinity of fluids in rare-metal systems



XCO₂ of fluids in rare metal systems



Bayan Obo: CO₂ from carbonate dissolution (Smith & Henderson, 2000)



calcite

apatite

Thank You



Justin Hoyle



Yonggang Feng

